

13 Abstract

14 Effluent from anaerobic membrane bioreactor (AnMBR) contains ammonia and would require
15 post-polishing treatment before it can be disinfected by chlorine. However, additional post-
16 treatment steps to remove nutrients offset the energetic benefits derived from anaerobic
17 fermentation. The use of chlorine or ozone also promotes concerns associated with disinfection
18 byproducts. This study evaluates UV/H₂O₂ as a potential strategy suited for the removal of
19 pharmaceutical compounds as well as antibiotic resistant bacteria (ARB) and antibiotic
20 resistance genes (ARGs) from AnMBR effluent. Our findings indicate that 10 mg/L H₂O₂ and
21 61.5 mJ/cm² of UV fluence are able to achieve a 4-log removal of both *Escherichia coli* PI7 and
22 *Klebsiella pneumoniae* L7. However, a higher fluence of 311 mJ/cm² with the same amount of
23 H₂O₂ would be required to achieve > 90% removal of atenolol, carbamazepine and estrone. The
24 removal of the pharmaceutical compounds was driven by the hydroxyl radicals generated from
25 H₂O₂, while UV exposure governed the inactivation of ARB and ARGs. UV/H₂O₂ increased
26 overall mutagenicity of the treated wastewater matrix but did not result in any changes to the
27 natural transformation rates. Instead, UV significantly reduced natural transformation rates by
28 means of DNA damage. Overall, UV/H₂O₂ could be the ideal final disinfection strategy for
29 AnMBR effluent without requiring additional post-treatment prior disinfection.

30
31
32
33
34

35 **Keywords:** Advanced oxidation processes, anaerobic membrane bioreactor, tertiary wastewater
36 treatment, contaminants of emerging concern, pharmaceutical compound degradation

37 1. Introduction

38 Agriculture irrigation currently accounts for approximately 70% of global freshwater
39 withdrawals (WWAP, 2015; 2019). A continuing trend in such unsustainable use of freshwater
40 will exacerbate water scarcity in arid regions like the Middle East and Africa. Reusing treated
41 wastewater for agricultural irrigation is a potential solution to alleviate the demand on freshwater
42 supplies. Regulatory agencies in Middle East and Africa recommend guidelines devised from
43 World Health Organization or International Organization for Standardization by listing
44 permissible limits of fecal coliforms, biochemical oxygen demand, total suspended solids and
45 turbidity in treated wastewater to be used for irrigation (Al-Jasser, 2011). However, emerging
46 contaminants such as the presence of residual antibiotic-resistant bacteria (ARB), antibiotic
47 resistance genes (ARG) and pharmaceuticals in treated wastewater effluent are not addressed in
48 the current guidelines. An increase in wastewater reuse could aid in the dissemination of
49 antibiotic resistance, while many chemicals of emerging concern have been shown to accumulate
50 in a variety of crops irrigated with treated wastewater (Goldstein et al., 2014; Malchi et al., 2014;
51 Wu et al., 2014). These observations arise due to an inconsistent removal of contaminants by
52 conventional wastewater treatment plants (WWTP) that rely on secondary treatment processes
53 (Verlicchi et al., 2012). There is therefore a need to enhance the removal of the aforementioned
54 contaminants through tertiary treatment, particularly by means of membrane filtration (Kim et
55 al., 2014) and chemical oxidation of wastewater (Okuda et al., 2008).

56 Microfiltration or ultrafiltration membrane bioreactors (MBR) are being considered as
57 tertiary treatment technologies as the coupling of membrane to activated sludge processes act as
58 an additional physical barrier that can improve effluent quality by removing particulates based
59 on size exclusion or by adsorbing particulates on the membrane biofoulant layer (Cheng and

60 Hong, 2017; Kim et al., 2014). To exemplify, MBRs achieve 1-3 log higher removal of ARG
61 compared to secondary activated sludge processes (Munir et al., 2011). The use of membrane
62 filtration also eliminates the need for clarifiers, hence decreasing the required land footprint and
63 facilitating installation of WWTP in a decentralized operation. Conventionally, MBR is operated
64 in aerobic mode but this is an energy intensive process. Furthermore, aerobic MBR generates a
65 large volume of solid wastes (i.e., sludge) concentrated with contaminants that would need to be
66 properly disposed. In recent years, anaerobic MBR (AnMBR) has been explored as a more
67 sustainable way to treat wastewater because of the recovery of methane as an energy source and
68 a low sludge volume production.

69 Regardless of the type of membrane bioreactor, a final disinfection step can be utilized to
70 complement the upstream treatment in inactivating bacteria and other contaminants, hence
71 augmenting the final cumulative log removal of contaminants. Chlorine is a common disinfectant
72 used in treated wastewaters. However, chlorination is ineffective in the removal of chemicals of
73 emerging concerns (CECs) (Choi et al., 2006; Lei and Snyder, 2007) and in controlling antibiotic
74 resistance (Fiorentino et al., 2015; Yuan et al., 2015). The use of chlorine also results in the
75 formation of hazardous disinfection byproducts (DBPs) (Bond et al., 2011; Huang et al., 2016;
76 Park et al., 2016; Yang et al., 2010) that are mutagenic and can trigger competent bacteria to gain
77 new genetic materials via natural transformation (Mantilla-Calderon et al., 2019). Furthermore,
78 AnMBR effluent is not suited to be disinfected directly with chlorine because AnMBR effluent is
79 inherently ammonia-rich and likely to result in the formation of nitrogenous DBPs upon
80 formation of chloramines (Shah and Mitch, 2012), which are highly toxic even in minute
81 concentrations (Hrudey, 2009; Nieuwenhuijsen et al., 2000). This implies that the use of
82 AnMBR to treat municipal wastewater for reuse would require further downstream nitrification

83 and denitrification biological systems prior to chlorine disinfection. However, the need for
84 downstream nutrient removal treatment increases the footprint of AnMBR, in turn making
85 AnMBR less appealing as a decentralized treatment technology. It also eliminates the possibility
86 of using high quality treated wastewater as a potential liquid fertilizer for food production.

87 In contrast with chlorine, UV is able to efficiently inactivate microorganisms due to
88 dimerization of pyrimidine bases in DNA when energy is absorbed in the UV-C range (Doetsch
89 et al., 1995; Sinha and Häder, 2002). UV can also impede possible horizontal gene transfer via
90 conjugation (Guo et al., 2015) and natural transformation (Augsburger et al., 2019; He et al.,
91 2019; McKinney and Pruden, 2012; Yoon et al., 2018) without generating disinfection
92 byproducts. However, UV alone via photolysis cannot proficiently remove CECs, hence there is
93 a need to supplement an oxidant such as H_2O_2 which has been shown to enhance the removal of
94 pharmaceutical compounds in combination with UV (Kim et al., 2009; Lee et al., 2016; Miklos
95 et al., 2018). UV/ H_2O_2 advanced oxidation process (AOP) is therefore proposed as a treatment
96 option for the removal of CECs from ammonia-rich AnMBR effluent. Ammonia is ~97%
97 transparent to the monochromatic light emitted by low-pressure UV lamps (254_{nm}) (Ling et al.,
98 2011; 2013), and hence the UV would be transmitted through the water matrix effectively to
99 inactivate ARG and ARB. UV can also react with H_2O_2 to generate hydroxyl radicals that
100 degrade pharmaceutical compounds. To the best of our knowledge, UV/ H_2O_2 has not been
101 demonstrated for the treatment of AnMBR effluent.

102 This study therefore aims to demonstrate the efficacy of UV/ H_2O_2 in removing both
103 biological and chemical contaminants from an ammonia-rich effluent collected post-AnMBR.
104 The AnMBR effluent was spiked with two pathogenic ARB (*Escherichia coli* PI-7 and
105 *Klebsiella pneumoniae* L7) and their associated plasmid-borne ARGs (*bla_{NDM-1}* and *bla_{CTX-M-1}*) as

106 well as three pharmaceutical compounds (estrone, carbamezapine, and atenolol) commonly
107 detected in wastewater effluent. The removal of these contaminants was monitored upon
108 exposure to UV/H₂O₂ so as to determine the needed UV fluence and H₂O₂ concentration to
109 achieve a 90% removal of the chemical contaminants and 4-log removal of the ARBs. Changes
110 in mutagenicity level and natural transformation frequencies were also monitored to determine if
111 UV/H₂O₂ would provide a safe means of disinfecting the wastewater.

112

113 2. Materials and methods

114 2.1 Anaerobic MBR (AnMBR) setup and parameters

115 The AnMBR operated in this study was of similar configuration to that operated in an earlier
116 study (Cheng et al., 2019). Briefly, the AnMBR was filled with cylindrical ceramic packing
117 medium of 1.5 cm diameter, with a 2 L working volume. Sludge was originally seeded with a
118 mixture of camel manure and anaerobic sludge from a wastewater treatment plant in Riyadh,
119 Saudi Arabia. The reactor was fed with municipal wastewater collected from KAUST
120 wastewater treatment plant and operated at 35 °C (mesophilic conditions) and pH 7. Prior to
121 feeding into the AnMBR, the influent stream is added with sodium acetate and starch to adjust
122 the COD to approximately 800 mg/L. The organic loading rate (OLR) was hence maintained at
123 approximately 0.69 g COD L⁻¹ day⁻¹. External cross-flow flat sheet polyvinylidene difluoride
124 (PVDF) microfiltration (MF) membrane (GE Osmonics, Minnetonka, MN, USA) was connected
125 to the anaerobic fermentation tank to form AnMBR. The membranes had a nominal pore size of
126 0.3 µm. The flux was maintained at approximately 7 L/m²/h (LMH) with an average hydraulic
127 retention time (HRT) of 27.8 h. No sludge was wasted throughout the study. Biogas was used to

128 scour the membrane surface at a gas sparging rate of 300 mL/min and the total recirculation rate
129 of AnMBR was maintained at 600 mL/min. The AnMBR has an average 96% COD removal
130 throughout the study and produced an average 219 mL CH₄ per g COD. AnMBR effluent was
131 collected daily and stored in sterile containers placed at 4 °C prior to use for inactivation
132 experiments.

133 2.2 Effluent sample preparation

134 Fresh stocks of atenolol, carbamazepine, and estrone, each of 10 mg/mL, were prepared less than
135 7 d prior to UV/H₂O₂ experiments. Carbamazepine and atenolol were dissolved in the same vial
136 containing methanol as solvent, while 1:1 methanol-water (v/v) was used to dissolve estrone.
137 AnMBR effluent was further filtered through 0.45 µm filter prior to spiking in with tested
138 contaminants. The effluent was spiked with each chemical to achieve final individual
139 concentrations of 50 µg/L. The spiked concentrations for these chemicals are 223-, 41- and 37-
140 fold greater than concentrations reported in wastewater influent for estrone, atenolol, and
141 carbamazepine, respectively (Krzeminski et al., 2019; Subedi and Kannan, 2015). The chemicals
142 were spiked at high concentrations to facilitate detection based on protocols described in section
143 2.4.

144 Effluents were also spiked with ARB and ARG. To prepare the ARB, a single colony of *E. coli*
145 PI-7 and *K. pneumoniae* L7, isolated and characterized in earlier studies (Mantilla-Calderon et
146 al., 2016; Toh et al., 2017), were individually inoculated into sterile 50 mL LB broth
147 supplemented with either 10 µg mL⁻¹ of meropenem (*E.coli* PI-7) or 8 µg mL⁻¹ of ceftazidime (*K.*
148 *pneumoniae* L7). These ARB were used because they are carbapenem-resistant and extended
149 spectrum beta-lactamase-producing Enterobacteriaceae, and are listed as priority pathogens of

150 concern by World Health Organization (WHO, 2017). The culture was incubated overnight at 37
151 °C and 200 rpm, and subsequently centrifuged at 5000 g for 10 min. The pelleted cells were
152 washed with 1X PBS, recentrifuged at 5000 g for 10 min, and the procedure was repeated twice.
153 The washed cells were then concentrated and adjusted to a stock concentration of ca. 10^{10}
154 cells/mL based on OD_{600nm} readings, and subsequently spiked into the AnMBR effluent. The
155 final cellular concentrations were ca. 10^9 cells/mL for the disinfection assays. To prepare ARG,
156 *bla_{NDM-1}* and *bla_{CTX-M-15}* genes, responsible for meropenem and ceftazidime, respectively, were
157 amplified from plasmids extracted from *E. coli* PI-7 and *K. pneumoniae* L7, respectively, using
158 the primers in Table S1. After running a PCR, stock concentrations of 10^{14} copies/mL of each
159 gene were obtained and subsequently spiked into the AnMBR effluent prior to AOP treatment
160 for final concentrations of 10^9 copies/mL. All disinfection experiments described in section 2.3
161 were conducted in reactors containing the CECs, ARB, and ARG.

162 2.3 Disinfection experiments for AnMBR effluent

163 UV disinfection was performed using an 8 W monochromatic UVP lamp emitting light in the
164 germicidal UV-C (254 nm) wavelength (Analytik Jena, Upland, CA). The samples were placed
165 14.5 cm from the lamp, which resulted in 0.55 mW/cm^2 of average irradiance at 254 nm,
166 measured with an SU-100-SS UV sensor (Apogee Instruments, Santa Monica, CA). The emitted
167 irradiance was calculated by multiplying the fluence rate by exposure time (Bolton and Linden,
168 2003). Microcosms used for UV-C_{254nm} experiments were placed within 250 mL glass beakers,
169 each with a diameter and height of 6.5 and 9.2 cm, respectively. Correction factors taking sample
170 depth and 254 nm transmittance into consideration were calculated to determine average UV
171 intensity throughout the sample. Only irradiances corrected for water quality and sample depth
172 are reported, with further details presented in Supporting Information Text 1. Spiked AnMBR

173 effluent samples were added into the reactors and mixed gently by a magnetic stir bar at 300
174 rpm, the temperature was maintained at 32 °C with a water bath. Before each experiment, the
175 UV lamp was warmed up for 5 min to ensure stable UV radiation fluence. The disinfection
176 experiments were carried out with 0, 10 mg/L, and 30 mg/L of H₂O₂, and up to 311 mJ/cm². Each
177 disinfection test was done in technical duplicates and with at least three biological replicates.

178 *2.4 Detection of chemical contaminants*

179 Degradation of atenolol, carbamazepine, and estrone was monitored throughout the UV/H₂O₂
180 treatment. Prior to detection, a solid-phase extraction (SPE) was run on 200 mL samples exposed
181 to 0, 52, 104, 207, and 311 mJ/cm² using an HLB Oasis column. The operating conditions for
182 SPE are listed in Table S2. The final eluate from SPE of each sample was subsequently
183 evaporated using the TurboVAP LV system at 65 °C with constant N₂ gas flow. After complete
184 evaporation, each sample was resuspended in 500 µL of methanol, which was used for further
185 detection and analysis with LC/MS.

186 The chromatographic separation of compounds in Agilent 1260 Infinity system was performed
187 with a C18 column with 5 µm pore size (150 x 20 mm dimension) that was further fitted with
188 guard column. A binary gradient of 0.1% ammonium hydroxide buffer, 4 mM ammonium
189 formate in 0.1 % formic acid (mobile phase A) and 4 mM ammonium formate in methanol
190 containing 0.1 % formic acid (mobile phase B) was used to provide the needed gradient for
191 compounds separation (Table S3). The flow rate was set at 800 µL/min for 20 min running time.
192 The mass spectrometer used was AB SCIEX QTRAP 5500. The analytes were detected using
193 multiple reaction monitoring (MRM) in negative mode. Detailed information on the MRM
194 transition for each chemical were provided in Table S4. The extraction efficiencies from the

195 AnMBR effluent were $92\% \pm 6\%$, $84\% \pm 4\%$, and $93\% \pm 4\%$ for atenolol, carbamazepine and
196 estrone, respectively. Standard curves for each CECs are shown in Figure S1.

197 2.5 Inactivation kinetics of ARB and ARG

198 Decay kinetics of the ARB were determined based on cell counts quantified over time through
199 LB agar. Log removal values (LRV) were calculated using the formula of $\log_{10}(N/N_0)$, where N
200 = CFU x dilution factor and N_0 = initial cellular concentration. The ARGs were quantified via
201 qPCR using specific target primers and probes, listed in Table S1. The qPCR assays were
202 conducted with an Applied Biosystems 7900HT Fast real-time polymerase chain reaction
203 (qPCR) System (Thermo Fisher Scientific, Carlsbad, CA). The qPCR standard curve was
204 produced as previously described (Cheng and Hong, 2017) by diluting in series to obtain
205 concentration ranging from 10^2 to 10^8 copies/ μL . Each 20 μL reaction volume consisted of 10
206 μL Taqman fast master mix, 1 μL of each primer (10 μM), 0.8 μL respective probe, 1 μL of
207 DNA template and 6.2 μL molecular grade H_2O . The thermal cycling profile was as follows: 50
208 $^\circ\text{C}$ for 2 min, 95 $^\circ\text{C}$ for 20 s, and 40 cycles of 95 $^\circ\text{C}$ for 1 s, 60 $^\circ\text{C}$ for 20 s. The standard curve
209 and no-template controls (NTC) were performed in triplicate, while test amplifications were
210 performed in duplicate. The amplification factor of the standards ranged from 1.89 to 2.04. All
211 the R -squared values of trendlines were > 0.94 and all the NTCs have C_q values > 34 .
212 Inactivation was presented as LRV using the formula of $\log_{10}(N/N_0)$, where N = copies/ μL
213 obtained at that sampling point and N_0 = the initial copies/ μL upon spiking.

214 2.6 Evaluation of mutagenicity levels

215 A modified Ames reversion assay in the tester strain *Salmonella enterica* serovar Typhimurium
216 was used to gauge toxicity of concentrated effluent as previously described (Maron and Ames,
217 1983). *Salmonella enterica* serovar Typhimurium TA100 and TA98 were used for mutagenicity

218 tests; this assay is considered the gold standard for assessing mutagenicity in bacterial systems
219 caused from substitution/point (TA 100) and frameshift (TA 98) mutations. A 400X organic
220 fraction concentrate stock of UV/H₂O₂-treated water was generated from solid phase extraction.
221 Subsequently, 50 μ L of stock was added to 950 μ L of the cellular suspension for the final
222 concentration of 20X treated water. Concentrates were generated from spiked AnMBR effluent
223 after 52, 104, 207, and 311 mJ/cm². The mutagenicity protocol was carried out by using with a
224 standardized ca. 10⁹ cells mL⁻¹ of the tester strain (determined based on OD_{600nm} measurements),
225 incubated at 37°C with 20X concentrate of treated water for 1 h. Subsequently, 100 μ L of the
226 standardized cell solution of each strain was plated in Vogel-Bonner (VB) minimal medium.
227 Both strains contain mutations preventing histidine synthesis, inhibiting growth in the minimal
228 medium. Colonies that are able to grow in this minimal medium agar plates represent a reversion
229 of these mutations due to exposure to a mutagen. The histidine revertant colonies were
230 enumerated by colony forming unit counts on agar plates after 72 h of incubation at 37 °C. The
231 results were expressed as fold changes, which was determined by normalizing the revertant
232 counts by the negative control of 50 μ L PPB for every mutagenicity assay. Each condition was
233 tested with biological triplicates.

234 2.7. Evaluation of extracellular DNA integrity

235 To evaluate the changes in the extracellular DNA integrity after the proposed disinfection
236 strategy, natural transformation assays were done using the reporter strain of *A. baylyi* ADP1
237 (BD413) along with its extracellular donor DNA (Rizzi et al., 2008). Briefly, the reporter strain
238 lacks a functional promotor for genes encoding for spectinomycin resistance (*aadA*) and
239 fluorescence (GFP). The donor DNA carrying the functional promotor was generated by PCR
240 using the pCLT cloning vector as template (Mantilla-Calderon et al., 2019). The primer set of 5'-

241 ATT ATT GAA TTC GGT AGA GCC GTT TAT GAA-3' forward and 5'-TTT GCC CAC TAC
242 CTT GGT GAT-3', amplifies the necessary donor DNA (2.1 kbp), which following translocation
243 into the cell, can integrate into the reporter strain's genome via homologous recombination
244 (Mantilla-Calderon et al., 2019). Integration of the promoter carried by the donor DNA activates
245 the transcription of the gene fusion, restoring spectinomycin resistance and fluorescence by *A.*
246 *baylyi* ADP1 (Rizzi et al., 2008). In these assays, only the donor DNA was exposed to the
247 UV/H₂O₂ treatment, and the reporter strain remained unexposed to ensure the change in natural
248 transformation rates was strictly due to DNA integrity and not any potential damage to the
249 reporter strain. The donor DNA was spiked into AnMBR effluent at a final concentration of 2
250 µg/mL and exposed to irradiances of 0-311 mJ/cm² with H₂O₂ concentrations of 0, 10, and 30
251 mg/L. After the varied UV/H₂O₂ conditions, the donor DNA containing wastewater sample was
252 transferred to a 2-mL centrifuge tube containing ca. 10⁶ cells/mL of the reporter (*A. baylyi*
253 ADP1) in LB broth, and the assay tubes were then placed in a shaker at 37 °C for a 24 h duration
254 of transformation. Each set of tests were done with three sets of controls, namely one without
255 donor DNA and exposed to UV/H₂O₂, one with respective H₂O₂ but kept in the dark for the
256 given amount of time, and another without H₂O₂ and kept in the dark. To quantify transformation
257 frequencies, 100 µL from the final assay tubes were plated on LB agar containing 100 µg/L of
258 spectinomycin, 50 µg/L of kanamycin and 50 µg/L of rifampicin. The number of CFU that grew
259 on these agar plates represent the transformant counts. These numbers of CFU were divided by
260 the CFU on LB without spectinomycin but with kanamycin and rifampin, which represent the
261 total number of cell counts. Each test was normalized by the dark control without H₂O₂ to
262 compute fold-changes in natural transformation frequencies. Statistical differences were

263 determined by comparing the transformation frequencies at each irradiation level to those in the
264 dark using one-way ANOVA. All experiments were tested in biological triplicates.

265 *2.8 Water quality and parameters of MBR effluent*

266 Levels of COD, ammonia, and nitrate were measured using Hach kits based on manufacturer's
267 protocol, while turbidity was measured using a Hach turbidity meter (Loveland, CO, USA).
268 SUVA was determined by dividing absorbance at 254 nm, measured using a UV-3600 UV-Vis
269 spectrometer. Dissolved organic carbon (DOC) was determined using Shimadzu TOC-V total
270 organic carbon analyzer after passing water samples through a 0.45 μm filter. The pH of each
271 sample was measured using a Fisher Scientific Accumet AB150 benchtop meter. To determine
272 the effect of the overall water quality on hydroxyl radicals, nitrobenzene degradation was
273 measured, using high performance liquid chromatography (HPLC), throughout UV/H₂O₂
274 treatment. The presence of hydroxyl radicals with nitrobenzene will result in the degradation of
275 nitrobenzene to nitrophenols in a dose-dependent rate (Bhatia, 1975). The results are shown in
276 Figure S2 and the method is further detailed in Supporting Information Text 2.

277

278 **3. Results**

279 *3.1 Degradation of chemical contaminants*

280 At 10 and 30 mg/L H₂O₂ for all three CECs analyzed in this study, there was a fluence-
281 dependent increment in percent degradation of each compound (Figure 1). In the case of
282 atenolol, a beta blocker used to treat high blood pressure, there was no significant difference in
283 the degradation percentage with increasing fluence in the absence of H₂O₂. However, once H₂O₂
284 is added, hydroxyl radical formation significantly increased the degradation of atenolol
285 significantly after just 52 mJ/cm² albeit at 30 mg/L of H₂O₂ (Figure 1A). With increasing UV

286 fluence $>104 \text{ mJ/cm}^2$, no difference in degradation percentage was observed when H_2O_2 was
287 increased from 10 to 30 mg/L. Average percent degradation at 311 mJ/cm^2 was also similar to
288 207 mJ/cm^2 for equal concentration of H_2O_2 . The maximum degradation of atenolol was 97% at
289 311 mJ/cm^2 and 30 mg/L H_2O_2 , but this percentage was not significantly different from the 96%
290 achieved at 311 mJ/cm^2 and 10 mg/L H_2O_2 (Table S5). For carbamazepine, an antiepileptic drug,
291 a significant improvement compared to 0 mg/L H_2O_2 was detected at 10 mg/L H_2O_2 after 207
292 mJ/cm^2 of UV exposure. Meanwhile a significant improvement in degradation upon increasing
293 H_2O_2 concentration from 0 mg/L to 30 mg/L was achieved at 52 mJ/cm^2 onwards. The maximum
294 degradation percentage of carbamazepine was 98%, achieved at 30 mg/L and 311 mJ/cm^2 but
295 this value was not different from the 97% achieved at 311 mJ/cm^2 and 10 mg/L H_2O_2 (Table S6).
296 The addition of H_2O_2 significantly increased the degradation of estrone, a female sex hormone,
297 even at the lowest irradiance of 52 mJ/cm^2 . Thereafter, a maximum estrone degradation achieved
298 with UV/ H_2O_2 was 92% at 30 mg/L H_2O_2 and 311 mJ/cm^2 (Table S7). The significance of the
299 addition of H_2O_2 is highlighted by the changes in the rate constants, assuming a second-order
300 reaction for all H_2O_2 concentrations. The k -values increased by 2.8, 2.0, and 1.39-fold for
301 atenolol, carbamazepine, and estrone, respectively, when increasing the H_2O_2 concentration from
302 10 to 30 mg/L at 311 mJ/cm^2 exposure (Table S8).

303

304 3.2 Inactivation of ARB

305 The inactivation of ARB *E. coli* PI7 and *K. pneumoniae* L7 in AnMBR effluent are shown in
306 Fig. 2. The slope determination and decay kinetics were calculated using the sampling points
307 between 0-52 mJ/cm^2 , where consistent logarithmic decay was observed throughout all samples
308 with R^2 values all above 0.88. The average slopes from the inactivation curves of *E. coli* PI7

309 were -0.01, -0.12, and -0.13 log-removal per mJ/cm^2 for 0, 10 and 30 mg/L H_2O_2 , respectively.
310 The slopes were then used to calculate the irradiances needed for a 4-log inactivation of *E. coli*
311 PI7 in the AnMBR effluent, which were 35.7, 25.2, and 25.4 mJ/cm^2 for 0, 10, and 30 mg/L
312 H_2O_2 . The addition of 10 mg/L significantly increased the inactivation rates of *E. coli* PI7 (Fig.
313 2A) based on slope comparisons (P-value = 0.02), however further increasing H_2O_2
314 concentration to 30 mg/L did not have a significant effect (P-value = 0.54). This same pattern
315 can be seen when comparing the first-order rate constants. The average k -values for 0, 10 and 30
316 mg/L throughout the same 0-52 mJ/cm^2 period are 2.22×10^{-1} , 2.77×10^{-1} , $3.82 \times 10^{-1} \text{ cm}^2/\text{mJ}$
317 (Table S9). Following logarithmic decay, there was a disparity observed in the residual cellular
318 concentration of *E. coli* PI7 (from 104-311 mJ/cm^2); being 2-log less when spiked with 10 mg/L
319 compared to 0 mg/L (Fig. 2A).

320 The inactivation of *K. pneumoniae* L7 (Fig. 2B) was less altered by the addition of H_2O_2
321 compared to *E. coli* PI7. Only the use of 30 mg/L H_2O_2 had a significant effect of inactivation
322 rates compared to 0 mg/L H_2O_2 (P-value = 0.02) when comparing slopes from 0-52 mJ/cm^2 .
323 However, the comparison slopes of 10 vs. 30 mg/L were not different. The decay kinetics from
324 0-52 mJ/cm^2 also slightly increased in the case of *K. pneumoniae* L7, whereby the k -values are
325 2.07×10^{-1} , 2.28×10^{-1} , $2.47 \times 10^{-1} \text{ cm}^2/\text{mJ}$ (Table S9). The required doses for a 4-log
326 inactivation are 37.7, 33.1, and 31.6 mJ/cm^2 when using 0, 10, or 30 mg/L H_2O_2 , respectively.
327 Again, there was a disparity in average residual *K. pneumoniae* L7 cell concentrations from 104-
328 311 mJ/cm^2 , which differed by 0.9-log between 0 and 10 mg/L H_2O_2 (Fig. 2B).

329

330 3.3 Inactivation of extracellular DNA

331 Degradation of extracellular DNA associated with ARGs was quantified using qPCR (Fig. 3).

332 The ARGs were spiked at 10^6 copies/ μ L; the detection limit was 10^2 copies/ μ L. Degradation of
333 ARGs was governed by UV exposure rather than the addition of H_2O_2 to UV. Without H_2O_2 ,
334 there is a clear positive UV-dose dependent relationship and the degradation of *bla_{CTX-M15}* (Fig.
335 3A), with a slope of 0.008 log-removal per mJ/cm^2 and an R^2 of 0.97 from 0-311 mJ/cm^2 and a
336 k -value of 2.43×10^{-2} cm^2/mJ when normalized by irradiance (Table S10). Without H_2O_2 , there
337 was a maximum removal of 2.7-log at 311 mJ/cm^2 which was different to that achieved at 104
338 mJ/cm^2 of exposure but not to that achieved at 207 mJ/cm^2 (Table S11). The addition of 10 mg/L
339 H_2O_2 did not significantly increase degradation of *bla_{CTX-M15}* at any irradiance, however, at 311
340 mJ/cm^2 there was an average degradation of 3.3-log, which is a 0.6-log more degradation
341 compared to 0 mg/L H_2O_2 . Interestingly, at 30 mg/L H_2O_2 there was no statistical difference at
342 any irradiance when comparing to 0 mg/L at any of the tested fluence as a pairwise comparison.
343 Overall the maximum addition of H_2O_2 (30 mg/L) resulted in just a 1.2-fold increase in the rate
344 constant k over the 311 mJ/cm^2 , compared to 0 mg/L H_2O_2 (Table S10).

345 In the degradation of *bla_{NDM-1}*, the same UV concentration-dependent response was noted (Fig.
346 3B). At 0 mg/L H_2O_2 , there was an increment in degradation with increase in UV dose following
347 an average slope of 0.006 log-removal per mJ/cm^2 and an R^2 of 0.93 from 0-311 mJ/cm^2 . A
348 maximum removal of 2.1-log was achieved at 311 mJ/cm^2 , however there was no significant
349 increase in degradation after 104 mJ/cm^2 (Table S12). With the addition of 10 mg/L H_2O_2 , there
350 was no significant difference at any irradiance compared to 0 mg/L apart from 311 mJ/cm^2 ,
351 where average degradation was increased to 2.9-log. Further increasing the H_2O_2 to 30 mg/L did
352 not result in improvement in LRV compared to that at 10 mg/L nor 0 mg/L at each respective
353 fluence. Lack of statistical significance in the changes of LRV as a result of increasing H_2O_2
354 concentration indicates that the removal of extracellular DNA was more governed by UV fluence

355 rather than H₂O₂ concentrations. The rate of that removal over the 311 mJ/cm² exposure,
356 represented by k increased slightly from 2.28×10^{-2} (0 mg/L H₂O₂) to 3.20×10^{-2} (10 mg/L
357 H₂O₂) and 3.34×10^{-2} (30 mg/L H₂O₂) as shown in Table S10.

358

359 *3.4 Mutagenicity effects of UV H₂O₂-treated water*

360 *Salmonella enterica* serovar Typhimurium TA98 and TA100 were used as mutagenicity assays
361 to assess possible frameshift and point mutations, respectively, using concentrated water samples
362 at each level of treatment (Fig. 4). The initial spiked effluent, prior to treatment showed no
363 difference to the negative control for either strain. Treated effluent without H₂O₂ did not illicit a
364 mutagenic response at any irradiance compared to the no-treatment sample in TA 100 (Fig. 4A).
365 There was however a significant increase in mutagenicity at 10 and 30 mg/L H₂O₂ starting at 104
366 mJ/cm² (Table S13). The maximum mutagenic response of 1.7-fold was observed after 311
367 mJ/cm² and 30 mg/L H₂O₂ in TA100. Furthermore, our results found no significant toxicity
368 increment due to UV/H₂O₂ treatment for all tested conditions using TA98 (Fig. 4B, Table S14).

369

370 *3.5. Transformability of DNA*

371 Although there was increase in mutagenicity associated with the treated wastewater matrix
372 (section 3.4), there was likewise a positive UV dose-response in DNA damage that further
373 prevented successful natural transformation of extracellular DNA that are present (Fig 5).
374 Without H₂O₂, the natural transformation frequency reported at 311 mJ/cm² UV exposure was
375 0.3-fold compared to dark control, which was significantly different to all other irradiances
376 without H₂O₂. The addition of 10 mg/L H₂O₂ resulted in a 0.2-fold induction that was not
377 different to UV alone at 311 mJ/cm² (Table S15). The addition of 30 mg/L H₂O₂ resulted in a

378 significant difference in transformation fold-induction to 0 mg/L H₂O₂ at every irradiance (Fig.
379 5). At the maximum tested irradiance, there was a 0.05-fold induction in natural transformation,
380 or a 1.34 log reduction compared to the dark controls.

381

382 3.6. Water quality

383 Both ammonia and COD were oxidized by H₂O₂, and the oxidation increased with UV fluence
384 (Table 1). The ammonia concentration was slightly reduced when exposed to 52-311 mJ/cm²,
385 although the difference was insignificant compared to untreated sample. The COD level reduced
386 by 72% after UV/H₂O₂. Both turbidity and SUVA also decreased after UV/H₂O₂ exposure on
387 AnMBR effluent, hence resulting in improved transmittance of the treated effluent. As AnMBR
388 effluent was derived after a microfiltration membrane, the initial transmittance of 72% suggests
389 suitability for UV/H₂O₂ treatment. However, a scavenging effect for the hydroxyl radicals
390 generated from H₂O₂ was observed for the AnMBR effluent (Fig. S2). At 10 mg/L H₂O₂, the
391 amount of nitrobenzene was significantly higher in the AnMBR effluent than that detected in
392 MilliQ water at > 207 mJ/cm² (P-value = 0.043), indicating a corresponding lower degradation
393 of nitrobenzene in AnMBR effluent than MilliQ could be due to the scavenging effect of organic
394 constituents in the treated wastewater.

395

396 4. Discussion

397 AnMBR is increasingly being considered as a sustainable technology for treating municipal
398 wastewaters. Recent demonstrations of several pilot-scale AnMBR in Spain, Italy, Korea (Foglia
399 et al., 2020; Martinez-Sosa et al., 2012; Shin et al., 2014) reported a LRV of > 4 -log for
400 coliforms (Foglia et al., 2020; Peña et al., 2019), which would not be sufficient to meet the

401 requirement for reuse purposes. Despite good LRV achieved by AnMBR alone, most guidelines
402 still require the tertiary treated wastewater to be disinfected prior to reuse and that certain
403 contaminants (e.g. dioxane) would have to be removed by no less than 0.5-log to meet the
404 California Code of Regulations (CCR) Title 22. AnMBR would therefore have to be coupled
405 with disinfection strategy in order for it to be used as one of the mainstream technologies to treat
406 municipal wastewater.

407 When choosing an appropriate alternative disinfection strategy, consideration of the (i)
408 practicality, (ii) efficacy of a broad range of contaminant removal, and (iii) minimal DBP
409 formation (and hence low mutagenicity and horizontal gene transfer rates) would be of great
410 importance. Ranking well in the three evaluated parameters would include UV/H₂O₂, O₃/H₂O₂,
411 and UV/Cl₂ (von Gunten, 2018). Strong oxidants such as ozone or OH radicals consume DBP-
412 forming precursors, lowering the risk arising from DBP formation by technologies such as
413 UV/H₂O₂ and O₃/H₂O₂. While O₃/H₂O₂ has been shown to be more economic for the removal of
414 CECs (Katsoyiannis et al., 2011), the main drawback for ozone-based treatment in bromide-
415 containing waters is the formation of bromate (Von Gunten and Hoigne, 1994). Albeit, in
416 ammonia-containing waters like that of AnMBR effluent quench HOBr, a bromate intermediate,
417 lowering bromate formation by up to 4-fold (Buffle et al., 2004; Soltermann et al., 2017). This
418 process, however, requires specific conditions including an oxidative treatment step
419 (chlorine/ozone) and minimizing OH-radicals (lowering pH) to ensure HOBr formation
420 (Soltermann et al., 2017) (Soltermann et al., 2017). The use of ozone would therefore require
421 optimization on-site and incur additional chemical costs for disinfection.

422 Although the transmittance of the AnMBR effluent was ca. 72% (Table 1) and that the
423 spiking of contaminants (e.g. ARB) at high concentrations would cause aggregation and light

424 scattering, the results still showed a relatively high inactivation efficiency of both ARB (5 LRV)
425 and ARG (2 LRV). This inactivation is mainly contributed by UV rather than H₂O₂, previously
426 demonstrated in (Pablos et al., 2013), where UV-C treatment with and without H₂O₂ gave the
427 same kinetic constant (2.30 min⁻¹). The inactivation efficiency achieved in this study agrees with
428 that reported by previous studies which saw 6 LRV of ARB in filtered wastewater after 3 min
429 (Serna-Galvis et al., 2018), and 1.5-3.6 LRV for ARG after 124mJ/cm² of UV-C/H₂O₂ treatment
430 with 10 mg/L H₂O₂ in pure water after 125 mJ/cm² (Yoon et al., 2018). Our results show that
431 particularly at higher exposure, there was a significant increase in total ARB inactivation.
432 Following the addition of H₂O₂, there was an improvement in ARB inactivation, which agrees
433 with another study that also saw a 1-3 log increase in bacterial inactivation in humic-containing
434 surface waters at > 311 mJ/cm² (Alkan et al., 2007). However, this study observed slightly lower
435 inactivation rates of ARGs with *k*-values of up to 3.34 x 10⁻² compared to that observed earlier,
436 namely *k* = 6.1 x 10⁻² (Yoon et al., 2018) and *k* = 1.1 x 10⁻¹ (Chang et al., 2017). Amplicon length
437 and water quality are likely the main contributors to this discrepancy among the *k*-values.

438 In contrast, UV alone is not that effective against CECs and would require the addition of
439 H₂O₂ for the purpose of removing pharmaceutical compounds (Kim et al., 2009). H₂O₂ is
440 photolyzed to form hydroxyl radicals which react non-selectively with organic and inorganic
441 compounds present in the wastewater matrix (Stefan, 2017). Possible radical scavengers present
442 in the water effluent limit the applicability of AOP, and generally wastewater with residual
443 organics would require higher H₂O₂ concentrations to achieve effective degradation of CECs
444 (Rosario-Ortiz et al., 2010). Furthermore, in this study, there was also scavenging by methanol,
445 which was introduced as the solvent for the CECs. After spiking, there was 0.19 mM methanol in
446 the AnMBR effluent which is oxidized and consumed relatively quickly by the OH radicals. This

447 would mean that higher UV irradiances might also be needed under these circumstances.
448 Nevertheless, with 207 mJ/cm² and 10 mg/L of H₂O₂, we were able to achieve 90% reduction of
449 the pharmaceutical compounds which was comparable to a study by Rosario Ortiz et al 2009,
450 where they observed a >90% degradation of atenolol and carbamazepine from tertiary effluent
451 with similar UV transmittance albeit with 20 mg/L H₂O₂ and 500 mJ/cm² (Rosario-Ortiz et al.,
452 2010).

453 Although this study only measured degradation and removal of the parent compound, it is
454 well established that transformation products form as a result of both oxidation (Lee and Von
455 Gunten, 2016) and photodegradation (Fatta-Kassinos et al., 2011). We observed that regardless
456 of the fate of these CECs, there was a slight increase in mutagenicity arising from point
457 mutations which was not observed prior to disinfection. Mutagenicity has been demonstrated to
458 increase natural transformation among competent bacteria (Augsburger et al., 2019; Mantilla-
459 Calderon et al., 2019). Despite components of UV/H₂O₂ promoting the transfer of genetic
460 components due to increase mutagenicity, there was no observed increase in the transformation
461 frequency (Figure 5). This is likely due to the damage of extracellular DNA by UV exposure,
462 which can preemptively inhibit successful HGT (Augsburger et al., 2019; Chang et al., 2017; He
463 et al., 2019) and has its positive implications for the potential spread of antibiotic resistance
464 and/or gaining of new functional traits. In this study, we only consider inactivation rates of
465 extracellular DNA exclusively, ignoring intracellular DNA which has been shown to have slower
466 inactivation kinetics (McKinney and Pruden, 2012). In (McKinney and Pruden, 2012), it was
467 demonstrated that after 100 mJ/cm² of UV-C irradiance of four different genes, there was an
468 average 0.14 LRV increase in removal of extracellular genes compared to its intracellular
469 equivalent, suggesting that our described ability to inactivate extracellular DNA and

470 detrimentally affect transformation may not happen at the same efficiency for intracellular DNA.
471 Similarly, McKinney and Pruden showed that the LRV achieved by UV-C varied among the four
472 genes, further suggesting that the size and nucleic acid composition of the DNA to be inactivated
473 by UV-C are important factors that can also affect the efficacy.

474 While the findings in this study suggest that UV/H₂O₂ is effective against CECs, ARB
475 and ARGs, these experiments were performed in a batch mode with significantly smaller
476 volumes and higher initial concentrations than that present in full-scale operations. While higher
477 CEC and cellular concentrations help with detection resolution, it also worsened water quality. In
478 real wastewater, relative rate constants of the dissolved fraction (CECs, dissolved organic matter
479 (DOM), cells, etc.) with either UV or OH-radical will determine the degradation rate and this
480 will heavily vary on a case-to-case basis. For this reason we see large variations in removal of
481 chemical contaminants in real wastewater (Krzeminski et al., 2019), possibly due to differing
482 concentrations of these contaminants. Furthermore, the effluent used in this study came from an
483 AnMBR equipped with at 0.3 µm nominal pore size microfilter, and was sequentially filtered by
484 a 0.45 µm syringe filter to minimize contamination risk for our culture-based quantification
485 methods. We acknowledge such procedures are equivalent to a double stage filtration process
486 that would lower the turbidity of the water and improve UV efficacy. Hence, translated
487 applications of UV/H₂O₂ to disinfect AnMBR effluent in full-scale systems would require
488 further evaluation. Finally, when translating for full-scale systems, batch mode disinfection such
489 as the one demonstrated in this study would be impractical due to the need for a large reactor
490 footprint. Thus, adjustments to the needed fluence and concentrations may be needed to achieve
491 the reported removal values. This can be done for instance, by allowing a suitable hydraulic flow

492 rate through a flow-through UV reactor to ensure sufficient contact time between the UV/H₂O₂
493 and the AnMBR effluent.

494

495 **5. Conclusions**

496 In this study, we demonstrated the use of UV/H₂O₂ as a potential strategy to reduce the levels of
497 CECs, ARB and ARGs that are present in the ammonia-rich AnMBR effluent. In such water
498 matrix, ARB and ARG inactivation was driven mainly by UV-C_{254nm} exposure. The addition of
499 H₂O₂ to UV disinfection was required to achieve efficient CECs degradation. While 10 mg/L
500 H₂O₂ significantly improved the degradation of trace CECs compared to UV alone, increasing
501 the H₂O₂ concentration to 30 mg/L did not cause significant improvement in the degradation
502 efficiency. Although there was an increase in mutagenicity of the treated waters likely due to the
503 products formed from the reaction between CECs and H₂O₂, there was no changes in the natural
504 transformation rates after UV/H₂O₂. These observations suggest that UV/H₂O₂ can act as an
505 efficient and safe disinfection strategy in removing both biological and chemical contaminants
506 from an ammonia rich AnMBR effluent. The proposed disinfection strategy also bypasses the
507 need to remove ammonia from the AnMBR effluent, hence allowing one to use such high-quality
508 treated effluent as liquid fertilizers for food production.

509

510 **6. Acknowledgement**

511 This study was supported by KAUST Center Applied Research Funding FCC/1/1971-32-01 and
512 Center of Excellence for NEOM Research flagship projects REI/1/4178-03-01 awarded to P.Y.-
513 H.

514

515

Journal Pre-proof

517 **Table 1.** Water quality parameters of the AnMBR effluent before and after spiking with all
 518 contaminants and during AOP treatment supplemented 10 mg/L H₂O₂.

	pH	Ammonia (mg/L NH₃-N)	COD (mg/L)	DOC (mg/L)	Turbidity (NTU)	SUVA (m⁻¹ mg⁻¹ L)	Transmittance (254nm)
AnMBR effluent	8.02 (± 0.04)	20.06 (± 1.46)	24.5 (± 3.39)	12.09 (± 2.46)	1.32 (± 0.25)	0.47 (± 0.20)	88 % (± 2.14)
After spiking	8.30 (± 0.13)	19.9 (± 0.4)	125.5 (± 0.57)	39.2 (± 3.42)	4.54 (± 0.35)	0.36 (± 0.09)	72 % (± 1.56)
52 mJ/cm²	8.77 (± 0.05)	16.2 (± 0.4)	77.6 (± 9.26)	23.4 (± 4.16)	4.59 (± 0.07)	0.35 (± 0.30)	87 % (± 6.94)
104 mJ/cm²	8.24 (± 0.16)	16.7 (± 1.9)	57.5 (± 3.54)	16.7 (± 2.91)	2.40 (± 1.03)	0.21 (± 0.09)	92 % (± 2.69)
207 mJ/cm²	8.31 (± 0.08)	14.9 (± 0.4)	41.5 (± 7.78)	14.4 (± 3.24)	2.93 (± 0.62)	0.24 (± 0.08)	92 % (± 3.13)
311 mJ/cm²	8.37 (± 0.06)	13.7 (± 3.6)	35 (± 8.49)	14.2 (± 5.14)	2.32 (± 0.62)	0.23 (± 0.20)	93 % (± 1.86)

519

520

521

522

523 **Figure legends**

524

525

526

527 **Figure 1.** Percent degradation of (A) atenolol, (B) carbamazepine, and (C) estrone as a function
528 of both fluence (i.e., 0, 52, 104, 207 and 311 mJ/cm²) and H₂O₂ concentrations (i.e., 0, 10 and 30
529 mg/L).

530

531 **Figure 2.** Inactivation curves of (A) *E.coli* PI7 and (B) *K. pneumoniae* L7 as a function of UV
532 exposure at varying H₂O₂ concentrations.

533

534 **Figure 3.** Inactivation of (A) bla_{CTX-M-15} and (B) bla_{NDM-1} genes as a function of fluence and at
535 varying H₂O₂ concentrations.

536

537 **Figure 4.** Mutagenic response (A) TA100 and (B) TA98 as a function of exposure to 20X
538 concentrated water samples collected after no treatment and various UV fluence with 10 mg/L
539 H₂O₂. Positive control denotes mutagenic response upon exposure to 100 µg/mL sodium azide.

540

541 **Figure 5.** Integrity of extracellular DNA damage assessed by natural transformation assay.

542 Extracellular DNA was exposed to various UV fluence with varying concentrations of H₂O₂.

543

544

545

546

547

548 **References**

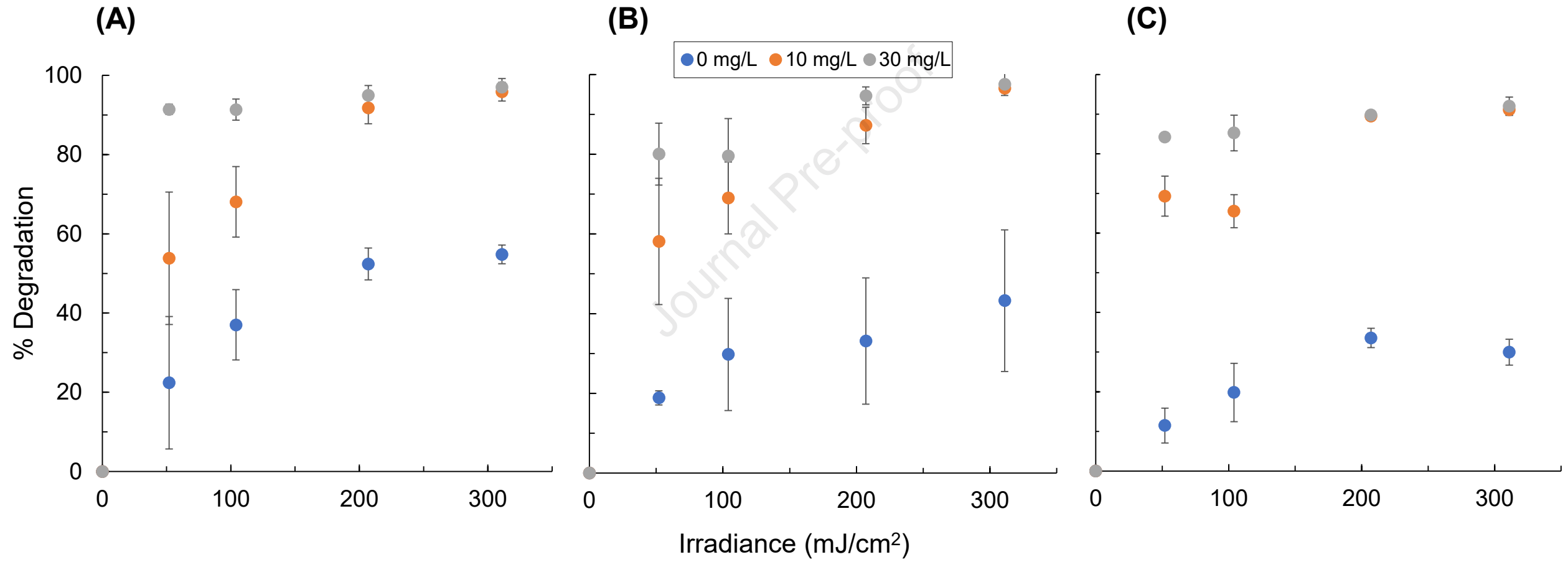
- 549
- 550 Al-Jasser, A. 2011. Saudi wastewater reuse standards for agricultural irrigation: Riyadh
551 treatment plants effluent compliance. *Journal of King Saud University-Engineering*
552 *Sciences* 23(1), 1-8.
- 553 Alkan, U., Teksoy, A., Atesli, A. and Baskaya, H.S. 2007. Efficiency of the UV/H₂O₂ process for
554 the disinfection of humic surface waters. *Journal of environmental science and health,*
555 *Part A* 42(4), 497-506.
- 556 Augsburger, N., Mantilla-Calderon, D., Daffonchio, D. and Hong, P.-Y. 2019. Acquisition of
557 Extracellular DNA by *Acinetobacter baylyi* ADP1 in response to Solar and UV-C254nm
558 Disinfection. *Environmental science & technology* 53(17), 10312-10319.
- 559 Bhatia, K. 1975. Hydroxyl radical induced oxidation of nitrobenzene. *The Journal of Physical*
560 *Chemistry* 79(10), 1032-1038.
- 561 Bolton, J.R. and Linden, K.G. 2003. Standardization of methods for fluence (UV dose)
562 determination in bench-scale UV experiments. *Journal of environmental engineering*
563 129(3), 209-215.
- 564 Bond, T., Huang, J., Templeton, M.R. and Graham, N. 2011. Occurrence and control of
565 nitrogenous disinfection by-products in drinking water—a review. *Water research* 45(15),
566 4341-4354.
- 567 Buffle, M.-O., Galli, S. and von Gunten, U. 2004. Enhanced bromate control during ozonation:
568 the chlorine-ammonia process. *Environmental science & technology* 38(19), 5187-5195.
- 569 Chang, P.H., Juhrend, B., Olson, T.M., Marrs, C.F. and Wigginton, K.R. 2017. Degradation of
570 extracellular antibiotic resistance genes with UV254 treatment. *Environmental Science*
571 *& Technology* 51(11), 6185-6192.
- 572 Cheng, H., Guan, Q., Villalobos, L.F., Peinemann, K.-V., Pain, A. and Hong, P.-Y. 2019.
573 Understanding the antifouling mechanisms related to copper oxide and zinc oxide
574 nanoparticles in anaerobic membrane bioreactors. *Environmental Science: Nano* 6(11),
575 3467-3479.
- 576 Cheng, H. and Hong, P.-Y. 2017. Removal of antibiotic-resistant bacteria and antibiotic
577 resistance genes affected by varying degrees of fouling on anaerobic microfiltration
578 membranes. *Environmental science & technology* 51(21), 12200-12209.
- 579 Choi, K.J., Kim, S.G., Kim, C.W. and Park, J.K. 2006. Removal efficiencies of endocrine disrupting
580 chemicals by coagulation/flocculation, ozonation, powdered/granular activated carbon
581 adsorption, and chlorination. *Korean Journal of Chemical Engineering* 23(3), 399-408.
- 582 Doetsch, P.W., Zastawny, T.H., Martin, A.M. and Dizdaroglu, M. 1995. Monomeric Base
583 Damage Products from Guanine, Adenine, and Thymine Induced by Exposure of DNA to
584 Ultraviolet Radiation. *Biochemistry* 34(3), 737-742.
- 585 Fatta-Kassinos, D., Vasquez, M.I. and Kümmerer, K. 2011. Transformation products of
586 pharmaceuticals in surface waters and wastewater formed during photolysis and
587 advanced oxidation processes—degradation, elucidation of byproducts and assessment
588 of their biological potency. *Chemosphere* 85(5), 693-709.
- 589 Fiorentino, A., Ferro, G., Alferez, M.C., Polo-López, M.I., Fernández-Ibañez, P. and Rizzo, L.
590 2015. Inactivation and regrowth of multidrug resistant bacteria in urban wastewater

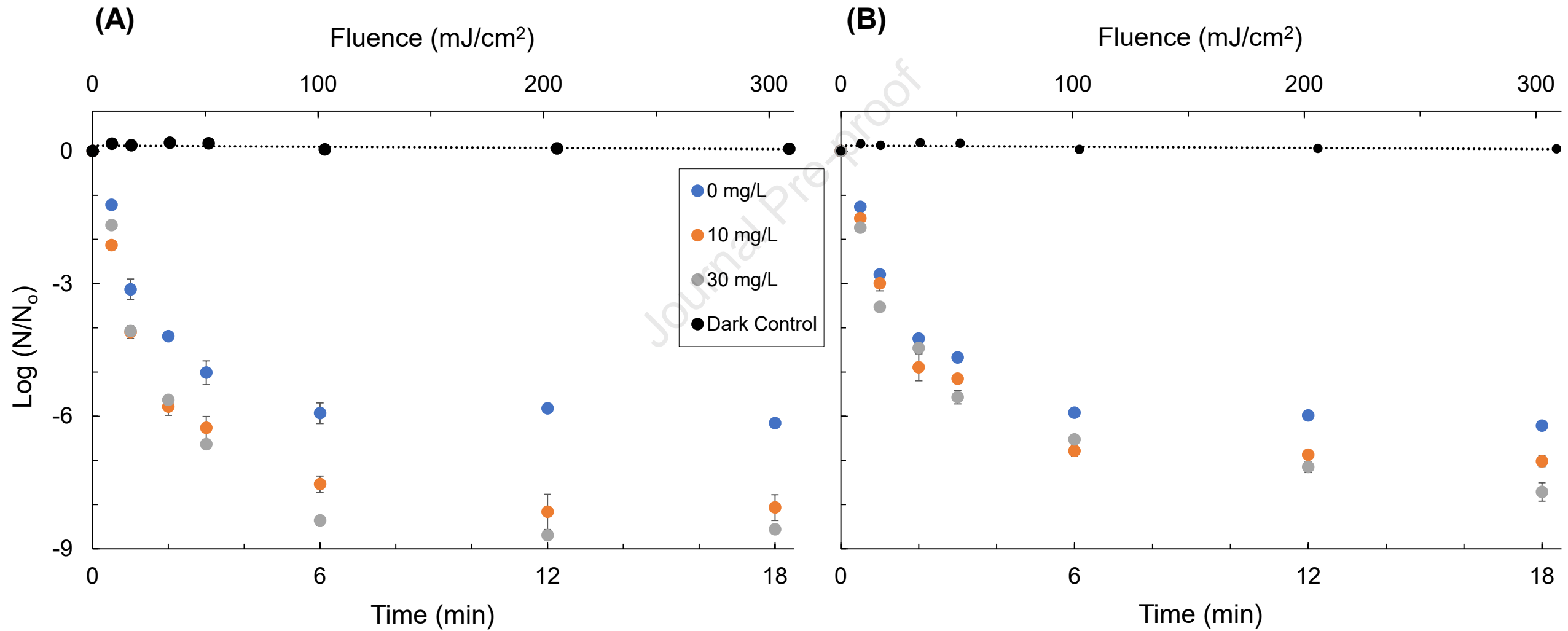
- 591 after disinfection by solar-driven and chlorination processes. *Journal of Photochemistry*
592 and *Photobiology B: Biology* 148, 43-50.
- 593 Foglia, A., Akyol, Ç., Frison, N., Katsou, E., Eusebi, A.L. and Fatone, F. 2020. Long-term
594 operation of a pilot-scale anaerobic membrane bioreactor (AnMBR) treating high salinity
595 low loaded municipal wastewater in real environment. *Separation and Purification*
596 *Technology* 236, 116279.
- 597 Goldstein, M., Shenker, M. and Chefetz, B. 2014. Insights into the uptake processes of
598 wastewater-borne pharmaceuticals by vegetables. *Environmental science & technology*
599 48(10), 5593-5600.
- 600 Guo, M.-T., Yuan, Q.-B. and Yang, J. 2015. Distinguishing effects of ultraviolet exposure and
601 chlorination on the horizontal transfer of antibiotic resistance genes in municipal
602 wastewater. *Environmental science & technology* 49(9), 5771-5778.
- 603 He, H., Zhou, P., Shimabuku, K.K., Fang, X., Li, S., Lee, Y. and Dodd, M.C. 2019. Degradation and
604 Deactivation of Bacterial Antibiotic Resistance Genes during Exposure to Free Chlorine,
605 Monochloramine, Chlorine Dioxide, Ozone, Ultraviolet Light, and Hydroxyl Radical.
606 *Environmental science & technology*.
- 607 Hrudey, S.E. 2009. Chlorination disinfection by-products, public health risk tradeoffs and me.
608 *Water research* 43(8), 2057-2092.
- 609 Huang, H., Wu, Q.-Y., Tang, X., Jiang, R. and Hu, H.-Y. 2016. Formation of haloacetonitriles and
610 haloacetamides and their precursors during chlorination of secondary effluents.
611 *Chemosphere* 144, 297-303.
- 612 Katsoyiannis, I.A., Canonica, S. and von Gunten, U. 2011. Efficiency and energy requirements
613 for the transformation of organic micropollutants by ozone, O₃/H₂O₂ and UV/H₂O₂.
614 *Water research* 45(13), 3811-3822.
- 615 Kim, I., Yamashita, N. and Tanaka, H. 2009. Performance of UV and UV/H₂O₂ processes for the
616 removal of pharmaceuticals detected in secondary effluent of a sewage treatment plant
617 in Japan. *Journal of Hazardous Materials* 166(2-3), 1134-1140.
- 618 Kim, M., Guerra, P., Shah, A., Parsa, M., Alaei, M. and Smyth, S. 2014. Removal of
619 pharmaceuticals and personal care products in a membrane bioreactor wastewater
620 treatment plant. *Water science and technology* 69(11), 2221-2229.
- 621 Krzeminski, P., Tomei, M.C., Karaolia, P., Langenhoff, A., Almeida, C.M.R., Felis, E., Gritten, F.,
622 Andersen, H.R., Fernandes, T. and Manaia, C.M. 2019. Performance of secondary
623 wastewater treatment methods for the removal of contaminants of emerging concern
624 implicated in crop uptake and antibiotic resistance spread: A review. *Science of the*
625 *Total Environment* 648, 1052-1081.
- 626 Lee, Y., Gerrity, D., Lee, M., Gamage, S., Pisarenko, A., Trenholm, R.A., Canonica, S., Snyder, S.A.
627 and Von Gunten, U. 2016. Organic contaminant abatement in reclaimed water by
628 UV/H₂O₂ and a combined process consisting of O₃/H₂O₂ followed by UV/H₂O₂:
629 prediction of abatement efficiency, energy consumption, and byproduct formation.
630 *Environmental science & technology* 50(7), 3809-3819.
- 631 Lee, Y. and Von Gunten, U. 2016. Advances in predicting organic contaminant abatement
632 during ozonation of municipal wastewater effluent: reaction kinetics, transformation
633 products, and changes of biological effects. *Environmental Science: Water Research &*
634 *Technology* 2(3), 421-442.

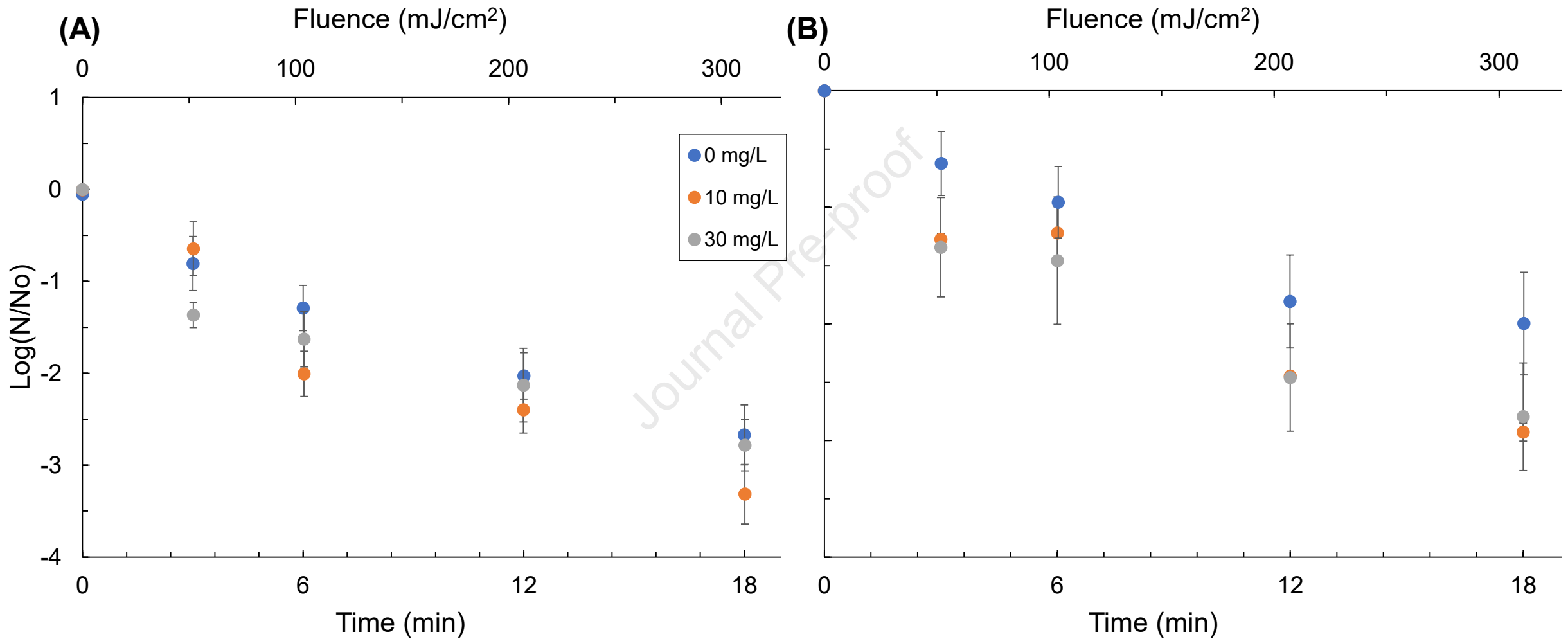
- 635 Lei, H. and Snyder, S.A. 2007. 3D QSPR models for the removal of trace organic contaminants
636 by ozone and free chlorine. *Water research* 41(18), 4051-4060.
- 637 Ling, T.L., Ahmad, M. and Heng, L.Y. 2011. Quantitative Determination of Ammonium Ion in
638 Aqueous Environment Using Riegler's Solution and Artificial Neural Network. *Sains*
639 *Malaysiana* 40(10), 1105-1113.
- 640 Ling, T.L., Ahmad, M. and Heng, L.Y. 2013. UV-vis spectrophotometric and artificial neural
641 network for estimation of ammonia in aqueous environment using cobalt (II) ions.
642 *Analytical Methods* 5(23), 6709-6714.
- 643 Malchi, T., Maor, Y., Tadmor, G., Shenker, M. and Chefetz, B. 2014. Irrigation of root
644 vegetables with treated wastewater: evaluating uptake of pharmaceuticals and the
645 associated human health risks. *Environmental science & technology* 48(16), 9325-9333.
- 646 Mantilla-Calderon, D., Jumat, M.R., Wang, T., Ganesan, P., Al-Jassim, N. and Hong, P.-Y. 2016.
647 Isolation and characterization of NDM-positive *Escherichia coli* from municipal
648 wastewater in Jeddah, Saudi Arabia. *Antimicrobial agents and chemotherapy* 60(9),
649 5223-5231.
- 650 Mantilla-Calderon, D., Plewa, M.J., Michoud, G., Fodelianakis, S., Daffonchio, D. and Hong, P.
651 2019. Water disinfection byproducts increase natural transformation rates of
652 environmental DNA in *Acinetobacter baylyi* ADP1. *Environmental Science & Technology*.
- 653 Maron, D.M. and Ames, B.N. 1983. Revised methods for the *Salmonella* mutagenicity test.
654 *Mutation Research/Environmental Mutagenesis and Related Subjects* 113(3-4), 173-215.
- 655 Martinez-Sosa, D., Helmreich, B. and Horn, H. 2012. Anaerobic submerged membrane
656 bioreactor (AnSMBR) treating low-strength wastewater under psychrophilic
657 temperature conditions. *Process Biochemistry* 47(5), 792-798.
- 658 McKinney, C.W. and Pruden, A. 2012. Ultraviolet disinfection of antibiotic resistant bacteria
659 and their antibiotic resistance genes in water and wastewater. *Environmental science &*
660 *technology* 46(24), 13393-13400.
- 661 Miklos, D.B., Remy, C., Jekel, M., Linden, K.G., Drewes, J.E. and Hübner, U. 2018. Evaluation of
662 advanced oxidation processes for water and wastewater treatment—A critical review.
663 *Water research* 139, 118-131.
- 664 Munir, M., Wong, K. and Xagorarakis, I. 2011. Release of antibiotic resistant bacteria and genes
665 in the effluent and biosolids of five wastewater utilities in Michigan. *Water research*
666 45(2), 681-693.
- 667 Nieuwenhuijsen, M.J., Toledano, M.B., Eaton, N.E., Fawell, J. and Elliott, P. 2000. Chlorination
668 disinfection byproducts in water and their association with adverse reproductive
669 outcomes: a review. *Occupational and environmental medicine* 57(2), 73-85.
- 670 Okuda, T., Kobayashi, Y., Nagao, R., Yamashita, N., Tanaka, H., Tanaka, S., Fujii, S., Konishi, C.
671 and Houwa, I. 2008. Removal efficiency of 66 pharmaceuticals during wastewater
672 treatment process in Japan. *Water Science and Technology* 57(1), 65-71.
- 673 Pablos, C., Marugan, J., van Grieken, R. and Serrano, E. 2013. Emerging micropollutant
674 oxidation during disinfection processes using UV-C, UV-C/H₂O₂, UV-A/TiO₂ and UV-
675 A/TiO₂/H₂O₂. *Water research* 47(3), 1237-1245.
- 676 Park, K.-Y., Choi, S.-Y., Lee, S.-H., Kweon, J.-H. and Song, J.-H. 2016. Comparison of formation
677 of disinfection by-products by chlorination and ozonation of wastewater effluents and
678 their toxicity to *Daphnia magna*. *Environmental Pollution* 215, 314-321.

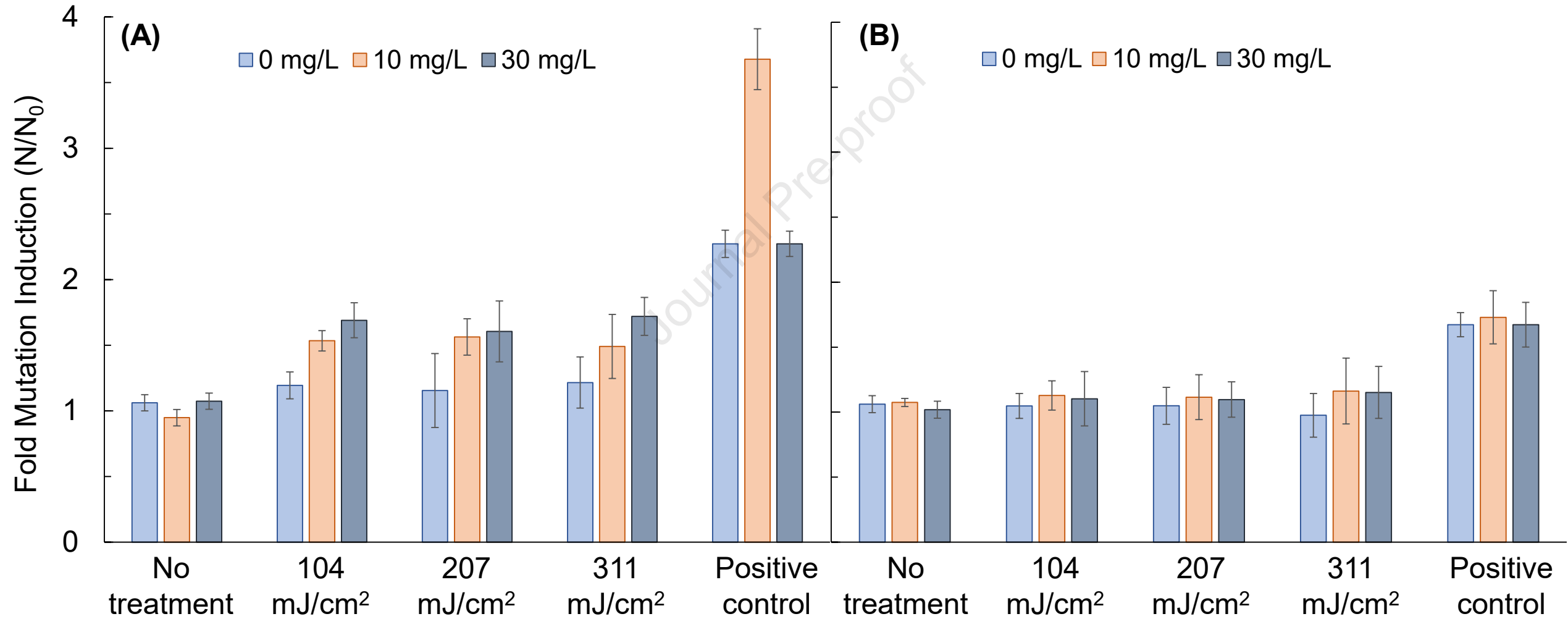
- 679 Peña, M., do Nascimento, T., Gouveia, J., Escudero, J., Gómez, A., Letona, A., Arrieta, J. and Fdz-
680 Polanco, F. 2019. Anaerobic submerged membrane bioreactor (AnSMBR) treating
681 municipal wastewater at ambient temperature: Operation and potential use for
682 agricultural irrigation. *Bioresource technology* 282, 285-293.
- 683 Rizzi, A., Pontiroli, A., Brusetti, L., Borin, S., Sorlini, C., Abruzzese, A., Sacchi, G.A., Vogel, T.M.,
684 Simonet, P. and Bazzicalupo, M. 2008. Strategy for in situ detection of natural
685 transformation-based horizontal gene transfer events. *Applied and environmental
686 microbiology* 74(4), 1250-1254.
- 687 Rosario-Ortiz, F.L., Wert, E.C. and Snyder, S.A. 2010. Evaluation of UV/H₂O₂ treatment for the
688 oxidation of pharmaceuticals in wastewater. *Water research* 44(5), 1440-1448.
- 689 Serna-Galvis, E.A., Salazar-Ospina, L., Jiménez, J.N., Pino, N.J. and Torres-Palma, R.A. 2018.
690 Elimination of carbapenem resistant *Klebsiella pneumoniae* in water by UV-C, UV-
691 C/persulfate and UV-C/H₂O₂. Evaluation of response to antibiotic, residual effect of the
692 processes and removal of resistance gene. *Journal of Environmental Chemical
693 Engineering*, 102196.
- 694 Shah, A.D. and Mitch, W.A. 2012. Halonitroalkanes, halonitriles, haloamides, and N-
695 nitrosamines: a critical review of nitrogenous disinfection byproduct formation
696 pathways. *Environmental Science & Technology* 46(1), 119-131.
- 697 Shin, C., McCarty, P.L., Kim, J. and Bae, J. 2014. Pilot-scale temperate-climate treatment of
698 domestic wastewater with a staged anaerobic fluidized membrane bioreactor (SAF-
699 MBR). *Bioresource Technology* 159, 95-103.
- 700 Sinha, R.P. and Häder, D.-P. 2002. UV-induced DNA damage and repair: a review.
701 *Photochemical & Photobiological Sciences* 1(4), 225-236.
- 702 Soltermann, F., Abegglen, C., Tschui, M., Stahel, S. and Von Gunten, U. 2017. Options and
703 limitations for bromate control during ozonation of wastewater. *Water research* 116,
704 76-85.
- 705 Stefan, M.I. (2017) *Advanced oxidation processes for water treatment: fundamentals and
706 applications*, IWA publishing.
- 707 Subedi, B. and Kannan, K. 2015. Occurrence and fate of select psychoactive pharmaceuticals
708 and antihypertensives in two wastewater treatment plants in New York State, USA.
709 *Science of the Total Environment* 514, 273-280.
- 710 Toh, B.E., Bokhari, O., Kutbi, A., Haroon, M.F., Mantilla-Calderon, D., Zowawi, H.M. and Hong,
711 P.-Y. 2017. Occurrence of extended-spectrum beta-lactamase Gram-negative bacteria
712 among various types of produce results in different level of ingestion risks.
- 713 Verlicchi, P., Al Aukidy, M. and Zambello, E. 2012. Occurrence of pharmaceutical compounds in
714 urban wastewater: removal, mass load and environmental risk after a secondary
715 treatment—a review. *Science of the total environment* 429, 123-155.
- 716 von Gunten, U. 2018. Oxidation processes in water treatment: are we on track? *Environmental
717 science & technology* 52(9), 5062-5075.
- 718 Von Gunten, U. and Hoigne, J. 1994. Bromate formation during ozonation of bromide-
719 containing waters: interaction of ozone and hydroxyl radical reactions. *Environmental
720 science & technology* 28(7), 1234-1242.
- 721 WHO 2017 List of bacteria for which new antibiotics are urgently needed. GENEVA, W. (ed),
722 Newsroom Detail.

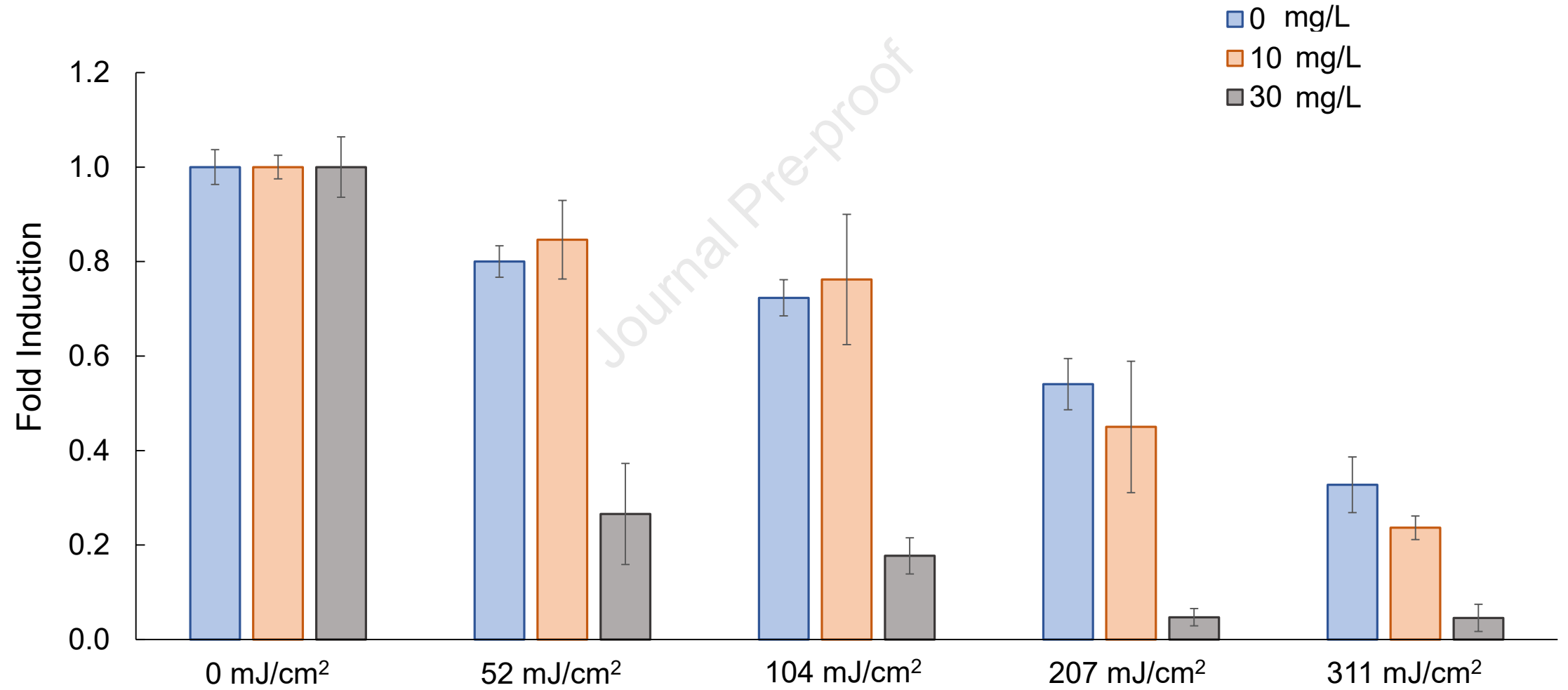
- 723 Wu, X., Conkle, J.L., Ernst, F. and Gan, J. 2014. Treated wastewater irrigation: uptake of
724 pharmaceutical and personal care products by common vegetables under field
725 conditions. *Environmental science & technology* 48(19), 11286-11293.
- 726 WWAP 2015. The Un World Water Development Report 2015. Water for a Sustainable World.
- 727 WWAP 2019 The United Nations World Water Development Report 2019: Leaving No One
728 Behind, UNESCO Paris.
- 729 Yang, X., Fan, C., Shang, C. and Zhao, Q. 2010. Nitrogenous disinfection byproducts formation
730 and nitrogen origin exploration during chloramination of nitrogenous organic
731 compounds. *Water research* 44(9), 2691-2702.
- 732 Yoon, Y., Dodd, M.C. and Lee, Y. 2018. Elimination of transforming activity and gene
733 degradation during UV and UV/H₂O₂ treatment of plasmid-encoded antibiotic
734 resistance genes. *Environmental Science: Water Research & Technology* 4(9), 1239-
735 1251.
- 736 Yuan, Q.-B., Guo, M.-T. and Yang, J. 2015. Fate of antibiotic resistant bacteria and genes during
737 wastewater chlorination: implication for antibiotic resistance control. *PLoS One* 10(3),
738 e0119403.
- 739
740
741











Highlights

- 10 mg/L H₂O₂ and 400 mJ/cm² UV removes 1-log of chemical contaminants from anaerobic effluent
- UV was the main factor to remove antibiotic resistant bacteria (ARB) and genes rather than H₂O₂
- > 5-log removal of ARB was achieved at > 400 mJ/cm² and > 10 mg/L H₂O₂
- UV/H₂O₂ significantly increased mutagenicity after 200 mJ/cm²
- Extracellular DNA removed by > 2-log and cannot be transformed into competent host
- Various aspects of water quality improved after disinfecting with UV/ H₂O₂

Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

Journal Pre-proof