

# 1 Seawater desalination using renewable energy: solar, geothermal, and wind

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## 9 10 **Abstract**

11 Globally, the Kingdom of Saudi Arabia (KSA) desalinates the largest capacity of seawater but  
12 through energy-intensive thermal processes such as multi-stage flash (MSF) distillation (>10  
13 kWh per m<sup>3</sup> of desalinated water). In other regions where fossil energy is more expensive and not  
14 subsidized, seawater reverse osmosis (SWRO) is the most common desalination technology but  
15 it is still energy-intensive (3-4 kWh/m<sup>3</sup>). Both processes therefore lead to the emission of  
16 significant amounts of greenhouse gases (GHGs). Moreover, MSF and SWRO technologies are  
17 most often used for large desalination facilities serving urban centers with centralized water  
18 distribution systems and power grids. While renewable energy (RE) sources could be used to  
19 serve centralized systems in urban centers and thus provide an opportunity to make desalination  
20 greener, they are mostly used to serve rural communities off of the grid. In the KSA, solar and  
21 geothermal energy are of most relevance in terms of local conditions. Our group is focusing on  
22 new desalination processes, adsorption desalination (AD) and membrane distillation (MD),  
23 which can be driven by waste heat or solar energy. A demonstration solar-AD facility has been  
24 constructed and a life cycle assessment showed that a specific energy consumption of <1.5  
25 kWh/m<sup>3</sup> is possible. An innovative hybrid approach has also been explored which would  
26 combine solar and geothermal energy using an alternating 12-hour cycle to reduce the  
27 probability of depleting the heat source within the geothermal reservoir and provide the most  
28 effective use of RE without the need for energy storage. This paper highlights the use of RE for  
29 desalination in KSA.

30  
31 *Keywords:* Renewable energy-driven Desalination; Combined solar-geothermal; Innovative  
32 desalination technologies; Environment, Saudi Arabia (KSA).

33 **1. Introduction**

34 Globally, the Kingdom of Saudi Arabia (KSA) desalinates the largest amount of water daily  
35 but, unfortunately, through energy-intensive thermal processes such as multi-stage flash  
36 distillation (MSF) in which the specific energy consumption is of about 10-16 kWh/m<sup>3</sup> [1-4]. In  
37 Europe, Australia and the USA, seawater reverse osmosis (SWRO) is the most common water  
38 desalination technology but is still energy-intensive (3 to 4 kWh/m<sup>3</sup>) compared to conventional  
39 drinking water supply sources such as surface or ground water treatment (<0.5 kWh/m<sup>3</sup>) [4]. Due  
40 to their energy demand, these processes lead to the emission of significant amounts of  
41 greenhouse gases (GHGs) when driven by on-grid power derived from conventional energy  
42 sources (e.g., about 1.0 kg CO<sub>2</sub>-equivalents (CO<sub>2</sub>-e) per kWh or 3.8 kg CO<sub>2</sub>-e per m<sup>3</sup> of  
43 desalinated water or 954 tons of CO<sub>2</sub>-e per day for the Sydney SWRO desalination facility with a  
44 capacity of 250,000 m<sup>3</sup>/day and a specific energy demand of 3.6 kWh/m<sup>3</sup> [5]. Moreover, because  
45 of their need for on-grid power, MSF and SWRO processes most often are limited to serving  
46 urban centers with centralized systems. While renewable energy (RE) sources such as solar,  
47 geothermal, and wind can still serve centralized systems in urban centers, they also provide an  
48 opportunity to make desalination *greener* and serve rural communities off of the grid or provide  
49 decentralized systems in urban areas, all with no *direct* GHG emissions. Australia has developed  
50 a national policy on requiring *energy compensation* for new desalination plants. For example,  
51 while the new Sydney SWRO plant takes its power from the grid, the energy demand of the  
52 facility is compensated by a newly developed wind farm in an offsite location with 67 turbines  
53 producing 132 MW of power. Presently, there are no large desalination plants being driven  
54 directly by RE, however, there are some small, stand-alone systems [2, 6-9]. In the KSA, solar  
55 and possibly geothermal energy are of most relevance in terms of local conditions.

56 Solar energy (SE) can be used for powering SWRO through photovoltaic (PV) cells which  
57 harness solar radiation. Alternatively, solar heat can be harvested by collectors, either  
58 concentrating (e.g., parabolic mirrors) or non-concentrating (e.g., solar ponds), for powering  
59 distillation [6]. The simplest approach is where the PV power generation and the desalination  
60 process are interfaced but not integrated; here, the problem is the variable diurnal pattern of  
61 sunlight. An integrated system requires an energy (heat) storage component for powering  
62 desalination during periods of low or no sunlight.

63 The King Abdulaziz City for Science and Technology (KACST) has a joint initiative with  
64 IBM, the National Initiative for Solar Desalination, in which 30,000 m<sup>3</sup>/day is being desalinated

65 by 10 MW of SE (Al-Lhafji plant); new improved PV cells from KACST are being coupled with  
66 SWRO membranes with improved materials from IBM [10, 11]. The Al-Lhafji solar water  
67 desalination plant will be, worldwide, the first high volume SWRO desalination plant powered  
68 by SE. To be most efficient, the plant design will profit from latest KACST/IBM UHCPV and  
69 Nanomembrane R&D results [10, 11].

70 Our group has developed a new low-energy desalination process, adsorption desalination  
71 (AD), which can be driven by waste heat or SE. A demonstration AD facility has been  
72 constructed on the KAUST campus, coupled with SE, and a life cycle assessment has been  
73 performed showing that a specific energy consumption of  $<1.5 \text{ kWh/m}^3$  is possible, with no  
74 *direct* GHG emissions. The inherent component of the AD process is silica gel, a very  
75 hydroscopic material, which adsorbs water from seawater which is then desorbed at a low  
76 temperature (55 to 85 °C) provided by SE, and later condensed free of salts. We are also  
77 performing research on solar-driven membrane distillation (MD) in which only water vapor from  
78 seawater is transported through a hydrophobic membrane by a temperature gradient as low as 40  
79 °C, which can be provided by SE. A recent breakthrough by our group has been the development  
80 of a new generation of MD membranes, opening the possibility of even lower temperature  
81 gradients. Details of these emerging technologies are presented in Section 6.

82 In the KSA, subsurface geothermal energy sources can potentially be tapped for powering  
83 desalination processes, suitable for AD, MD and multi-effect distillation (MED) ( $< 100 \text{ °C}$ ), and  
84 possibly even for MSF ( $> 100 \text{ °C}$ ). Geothermal energy can be harvested as wet or dry, and a heat  
85 exchanger can be used within a closed loop system. The WDRC is exploring an innovative  
86 hybrid approach which would combine solar and geothermal energy using an alternating 12-hour  
87 cycle to reduce the probability of depleting the heat source within the geothermal reservoir and  
88 provide the most effective use of RE without the need for energy storage.

89 This paper highlights the use of RE mainly solar, geothermal and wind energies for  
90 conventional and the innovative desalination technologies described with an emphasis on  
91 application in KSA.

92

## 93 **2. Desalination capacity and market in KSA**

94 KSA is the world's largest producer of desalinated water, accounting for 18 % of total global  
95 output and 41 % of the total production capacity of the Gulf Corporation Council (GCC) states  
96 [12]. The desalination capacity in KSA is approaching a daily water production of 10 million

97 cubic meters per day ( $\text{Mm}^3/\text{d}$ ) (Table 1) [12]. In addition to the existing capacity of  $9.8 \text{ Mm}^3/\text{d}$ , a  
98 further  $1.6 \text{ Mm}^3/\text{d}$  is under consideration [12, 13]. Most of the desalinated water in the KSA is  
99 still produced by thermal-based desalination processes, mainly the MSF process ( $5.6 \text{ Mm}^3/\text{d}$ ),  
100 but less energy-intensive processes such as MED and reverse osmosis (RO) are gaining ground.

101

### 102 **3. Desalination and energy**

103 The energy demand of desalination depends on a range of factors including recovery,  
104 pretreatment design (e.g., conventional vs. membrane filtration), the type of distillation process  
105 (e.g., MSF vs. MED) or SWRO membranes used (e.g., low energy membranes), the efficiency of  
106 pumps and motors, the type and efficiency of the energy recovery system installed (if any), and  
107 environmental conditions (e.g., feed water temperature). Energy demand also depends on the  
108 product water specifications. For example, employing a second SWRO pass for boron removal  
109 will increase the energy demand of the process [14]. Table 2 [15-18] summarizes the typical  
110 energy requirements of the main desalination processes and compares them to other water supply  
111 options [1, 5].

112 Modern SWRO plants can achieve a specific energy demand of  $<2.5 \text{ kWh}/\text{m}^3$  and a total  
113 energy demand of  $<3.5 \text{ kWh}/\text{m}^3$  by using state of the art equipment (such as pressure  
114 exchangers, variable frequency pumps and low-pressure membranes) and under favorable  
115 conditions (i.e., a low fouling potential, a temperature  $> 15 \text{ }^\circ\text{C}$ , a salinity  $< 35,000 \text{ ppm}$ ). The  
116 real energy demand may be higher under less favorable conditions. For example, the calculated  
117 specific energy demand of a state-of-the-art facility with a feed water salinity of  $40,000 \text{ ppm}$  and  
118 a temperature of  $20 \text{ }^\circ\text{C}$  (typical for Eastern Mediterranean seawater), a total recovery of  $41 \%$ ,  
119 and equipped with the most efficient energy recovery system, is approximately  $3.8 \text{ kWh}/\text{m}^3$  [15].  
120 An additional  $0.2\text{-}0.8 \text{ kWh}/\text{m}^3$  is required for pretreatment, waste water and sludge treatment  
121 (depending on the feed water quality), administration buildings and laboratories, post-treatment  
122 and drinking water pumping to supply network, which leads to a total energy consumption of  
123 about  $4\text{-}4.6 \text{ kWh}/\text{m}^3$  [19]. Ghaffour et al. [1] reported real data from two SWRO plants, using  
124 state of the art equipment, located in different sites. The total power consumption (including  
125 pretreatment and post-treatment) of a first one-pass SWRO plant (salinity =  $35,000 \text{ ppm}$ ) is  $3.8$   
126  $\text{kWh}/\text{m}^3$ , whereas a second two-pass SWRO plant (salinity =  $39,000 \text{ ppm}$ ) consumes  $4.25$   
127  $\text{kWh}/\text{m}^3$ . For example, the Spanish National Hydrological Plan assumes a total energy value of  $4$   
128  $\text{kWh}/\text{m}^3$  under the assumption that plants are equipped with state of the art technologies [20],

129 which is similar to the energy demands reported for other large SWRO projects around the  
130 world. Older or smaller SWRO plants without energy recovery may use up to 7 kWh/m<sup>3</sup> [15].

131 As the treatment and distribution of water by conventional means also require energy, the  
132 relative increase in energy demand should be considered in addition to the total demand of the  
133 process. The electrical energy demand of treating local surface water is typically between 0.2 and  
134 0.4 kWh/m<sup>3</sup>, compared to a specific energy demand of a modern SWRO plant of 3.5 kWh/m<sup>3</sup>  
135 under favorable conditions, resulting in the best case in a relative increase of 3.1–3.3 kWh/m<sup>3</sup> for  
136 seawater desalination. In locations where the water is transported over long distances, the  
137 relative increase of a local desalination plant may be much smaller – or even the best option as in  
138 the Perth example.

139 MSF distillation plants, which have an operating temperature up to 120°C, require about 250-  
140 330 MJ of thermal energy and 3-5 kWh of electrical energy for the production of one cubic  
141 meter of water (Table 2) [15]. MED plants, which operate at temperatures below 70°C, require  
142 145-390 MJ of thermal and 1.5-2.5 kWh of electrical energy per m<sup>3</sup> of water [15]. Although  
143 distillation processes require more energy than SWRO, they are still the first choice in countries  
144 of the Middle East (cf. section 2) for political, technical and economic reasons, such as difficult  
145 feed water conditions for SWRO plants (e.g., red tide events) and the availability of low cost  
146 energy. Dual-purpose co-generation facilities predominate in the region, which integrate MSF or  
147 MED distillation with power generation [21]. Because MSF and MED are capable of using ‘low  
148 value’ and ‘waste’ heat<sup>1</sup>, it is not straightforward to compare the total energy use of distillation  
149 with SWRO.

150 In a comparative life cycle assessment of different desalination processes it was concluded that  
151 the environmental impact of SWRO is one order of magnitude lower than the impact of thermal  
152 processes if they are operated with a conventional boiler, but comparable if the thermal processes  
153 are entirely driven by waste heat. MED was found to be more efficient than MSF and was also  
154 more energy efficient than RO in one evaluation under the assumption that waste heat is used.  
155 This can also be seen if only the electrical energy demand is compared (Table 2). The  
156 environmental impact of distillation processes can therefore be significantly reduced (by up to  
157 75%) if integrated into other industrial processes [22-24].

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<sup>1</sup> Waste heat is defined as heat that is released to the environment, such as steam leaving a backpressure turbine that can no longer be used to produce electricity. Low value heat is defined as heat of low temperature with little value for industrial processes, such as steam extracted from a condensing turbine, that could still be used to generate electricity, but that is sometimes wasted depending on practical circumstances such as electricity demand [15].

158

#### 159 **4. Climate change and GHG emissions from desalination**

160 As both the electrical and thermal energy used for the desalination of seawater is typically  
161 produced from fossil energy sources, a main environmental and public health concern of  
162 desalination is the release of air pollutants into the atmosphere, primarily GHGs (CO<sub>2</sub>), acid rain  
163 gases (NO<sub>x</sub>; SO<sub>2</sub>), and fine particulate matter. The emissions can result directly from the  
164 process, i.e., when fossil fuels are used to provide thermal heat in the form of steam for thermal  
165 desalination processes in power and water cogeneration plants, or indirectly when electricity is  
166 taken from the electricity grid to be used in the desalination process.

167 GHGs and air pollutant emissions generally depend on the fuel type, the efficiency of the  
168 power plant that produces the electricity for the desalination plant as well as the installed exhaust  
169 purification equipment at that plant. CO<sub>2</sub> emissions can be estimated with a high degree of  
170 certainty, as they depend mainly on the carbon content of the fuel (while carbon sequestration  
171 techniques are not widely in use yet). In order to take all relevant climate-change gases into  
172 account that arise from the combustion of fossil fuels (namely CO<sub>2</sub>, methane, nitrous oxide,  
173 perfluorocarbons, hydrofluorocarbons and sulphur hexafluoride, as specified in the Kyoto  
174 protocol), all climate change gas emissions can be expressed as CO<sub>2</sub>-equivalents (CO<sub>2</sub>-e), i.e.,  
175 the equivalent amount of CO<sub>2</sub> that would have the same global warming potential as the non-  
176 CO<sub>2</sub> emissions. CO<sub>2</sub>-e emission factors have been established as part of the international  
177 emission trading schemes. When electricity is taken from the grid, the energy mix of the  
178 respective grid must furthermore be taken into account. In the following, CO<sub>2</sub>-e emission  
179 estimates for SWRO plants will be provided assuming a specific energy demand of 4 kWh/m<sup>3</sup>.

180 For example, a grid average value of 1.16 kg CO<sub>2</sub>-e per kWh was used to calculate the  
181 emissions for the Gold Coast desalination project in Queensland, Australia [25]. For the  
182 Victorian SWRO plant in Melbourne, a grid average value of 1.31 kg CO<sub>2</sub>-e per kWh applies,  
183 which is the highest in the whole of Australia due to a high share of brown coal in the energy  
184 mix of Victoria [26].

185 For comparison, if one assumes an energy demand of 4 kWh/m<sup>3</sup> for the SWRO process, CO<sub>2</sub>-e  
186 emissions would be 4.6 kg/m<sup>3</sup> of desalinated water in the Queensland case and 5.2 kg/m<sup>3</sup> in the  
187 Melbourne case. The emissions would be even higher if one furthermore includes the distribution  
188 of the water, construction activities as well as CO<sub>2</sub> emissions associated with the use of materials  
189 and chemicals into the calculation. The real energy demand of the Queensland plant including all

190 operations and pumping within the existing water storage network is estimated to be 24.5 MW  
191 for a capacity of 125,000 m<sup>3</sup>/d, which equates to 4.7 kWh/m<sup>3</sup> (4.1 kWh/m<sup>3</sup> for desalination  
192 alone). The indirect GHGs emissions as a result of electricity use are estimated to be  
193 approximately 679 tons of CO<sub>2</sub>-e per day, or 5.43 kg of CO<sub>2</sub>-e per m<sup>3</sup> of desalinated water,  
194 which represents a 2% increase in emissions in the Gold Coast region [25].

195 The real energy demand arising from the Melbourne SWRO project was estimated to be 3,239  
196 tons of CO<sub>2</sub>-e per day which equates to 7.45 kg of CO<sub>2</sub>-e per m<sup>3</sup> of desalinated water. This value  
197 covers electricity used to drive the process and transfer the water, as well as emissions from the  
198 transportation of workforce, wastes and chemicals, from offsite waste decomposition, and  
199 embodied in the chemicals used during operation. Furthermore, if the energy used during  
200 construction of the project is added, amortized over the project life of 30 years, the energy  
201 demand amounts to 7.8 kg of CO<sub>2</sub>-e per m<sup>3</sup>. However, the construction process accounts for only  
202 4% of the total project emissions [26].

203 For Sydney and Perth, where other large desalination projects are located, emission factors for  
204 electricity from the grid are 1.06 and 0.98 kg CO<sub>2</sub>-e per kWh, respectively [27]. The electricity  
205 demand arising from the Sydney SWRO project with an initial capacity of 250,000 m<sup>3</sup>/d may  
206 result in emissions of 4.24 kg of CO<sub>2</sub>-e per m<sup>3</sup> or 1,060 tons of CO<sub>2</sub>-e per day. The Perth project  
207 has the lowest grid-specific emission factor and lowest reported energy demand of the Australian  
208 projects, resulting in the lowest CO<sub>2</sub>-e emissions of 3.43 kg/m<sup>3</sup> for the whole plant. The  
209 desalination-specific emissions amount to 325 t/d.

210 The cited examples all calculate the GHGs emissions for electricity purchased from the grid  
211 applying the grid-specific emission factor. It would also be possible to co-locate a SWRO plant  
212 next to an existing power plant, which would supply electricity 'over-the-fence' to avoid  
213 transmission losses. In Ashkelon, the SWRO process is driven by a natural gas-fired power plant  
214 on site, which supplies 50 MW of electricity to the desalination plant with a capacity of 330,000  
215 m<sup>3</sup>/d (3.6 kWh/m<sup>3</sup>). Applying a CO<sub>2</sub> emission factor for natural gas of 202 g CO<sub>2</sub>/kWh results in  
216 a very low emission factor of 0.73 kg CO<sub>2</sub> per m<sup>3</sup>. However, it could be argued that the  
217 electricity for the desalination plant could also serve other end-users, so that it would be more  
218 accurate to apply the grid-specific emission factor.

219 A worst case example for energy demand of desalination processes and air pollutant emissions  
220 are the Gulf countries which depend heavily on desalinated seawater from co-generation plants.  
221 In Kuwait, for instance, co-generation plants produce 90% of the national water supply and are

222 almost exclusively fired by heavy crude oils. Kuwait has the fourth largest seawater desalination  
223 capacity on a global scale after Spain and accounts for 6% of the worldwide production [5]. In  
224 2004, the plants generated 42 million MWh of electricity and 443 million m<sup>3</sup> of water, using 462  
225 million GJ of energy, which is 54% of the national fuel use. The corresponding CO<sub>2</sub> emission  
226 factors are 0.7 kg/kWh for electricity and 15.7 kg/m<sup>3</sup> for desalination. The total CO<sub>2</sub> emissions  
227 are approximately 19.000 t/d for 1.2 million m<sup>3</sup> of water per day. In Kuwait, desalination  
228 accounts for 10% of the national fuel use and hence the national emissions [28].

229

## 230 **5. The use of renewable energy resources in desalination**

231 Only about 1% of total desalinated water is currently based on energy from renewable sources.  
232 However, as renewables are becoming increasingly main stream and technology prices continue  
233 to decline, there is a large market potential for RE-powered desalination systems worldwide.  
234 Renewable technologies that are suited to desalination include solar thermal and PV, wind, and  
235 geothermal energy. As electricity storage is still a challenge, combining power generation and  
236 water desalination can also be a cost-effective option for electricity storage when generation  
237 exceeds demand [29].

238

### 239 *5.1 Solar desalination*

240 In the semi-arid and desert regions, solar energy is available in abundance with a typical solar  
241 irradiation of 2,200 to 2,400 kWh/m<sup>2</sup>.year. Depending on the type of receiver devices and the  
242 application temperatures, the collectable amount of energy of the collectors may vary from 600  
243 to 1,500 kWh/m<sup>2</sup>.year, which corresponds to efficiencies of 25% to 60%. For example, a  
244 selectively-coated flat-plate and stationary thermal collector in an equatorial region may have a  
245 solar thermal rating of 925 kW/m<sup>2</sup>.year with a thermal collection efficiency of 45% at 80 °C of  
246 application temperature, as compared with 1,100 to 1,200 kWh/m<sup>2</sup>.year for the desert regions.  
247 For higher application temperatures, e.g., up to 350 °C, the single or two axis concentrators are  
248 frequently used. As solar rays are optically concentrated onto a focused tubing with fluids  
249 flowing internally (either thermal oil or pressurized water), the application temperature attained  
250 is typically up to 300 to 400 °C. At such temperatures, the heat could be used directly in a co-  
251 generation plant for both power and low temperature steam extraction to power either the MSF  
252 or the MED desalination cycles. A well-designed concentrator system could have a thermal  
253 rating of 1,300 to 1,500 kWh/m<sup>2</sup>.year. However, one inherent disadvantage factor of solar



254 collectors in desert regions is the dust cover which may lower the thermal rating significantly if  
255 the collectors are unattended. The other drawback of a solar powered system is that it can only  
256 provide energy input during day light hours and, should a desalination facility need to operate  
257 continuously, large capacity thermal storage is needed. As a rule of thumb, a conventional silicon  
258 PV system would have an electricity rating of 110-120 kWh/m<sup>2</sup>.year. Thus, the collector area  
259 needed to produce 1 m<sup>3</sup>/day of water from a small RO plant (with a total specific energy  
260 consumption of 8 kWh/m<sup>3</sup>) is 26.5 to 28 m<sup>2</sup> of conventional PV collectors.

261

## 262 5.2 Desalination using wind energy

263 Wind power based desalination can be one of the most promising options for seawater  
264 desalination, especially in coastal areas with high wind potential. The electrical and mechanical  
265 power generated by a wind turbine can be used to power desalination plants, notably RO or  
266 distillation by mechanical vapor compression (MVC). Various small wind-based desalination  
267 plants have been installed around the world. As with SE, a drawback of wind desalination is the  
268 intermittence of the energy source. Possible combinations with other RE sources, batteries or  
269 other energy storage systems can provide smoother operating conditions [29]. The ENERCON  
270 Desalination System (EDS) for instance was specifically designed for combination with a wind  
271 energy converter. In contrast to conventional SWRO plants, the EDS has a flexible production  
272 capacity that can be continuously adjusted from 12.5 to 100 % of nominal capacity, thereby  
273 allowing for easy adjustment to changing power availability as caused by fluctuating wind  
274 conditions — or changing water demand [30].

275 Besides using wind energy *directly* for desalination, the electricity grid can also serve as a  
276 means for using wind energy (or other renewables) *indirectly*, i.e., by compensating or offsetting  
277 the electricity demand by desalination by renewables produced off-site. For example, the two  
278 Perth SWRO plants and the Sydney plant in Australia compensate their electricity demand by  
279 newly erected wind farms: For the first Perth plant, an associated 82 MW wind farm with 48  
280 turbines was erected 200 km north of the city, which will inject an expected 272 GWh of RE into  
281 the grid per year, from which the Perth plant purchases 185 GWh (68%) per year [18, 31]. The  
282 wind farm that has been purpose-built near Canberra to offset the energy use of the Sydney  
283 SWRO project will consist of 67 wind turbines and will provide 132 MW of energy to the grid,  
284 versus 42 MW required to operate the plant [32].

285 However, *offsetting* energy demand and GHGs emissions needs to be carefully evaluated to  
286 ensure that these projects are in fact GHGs-neutral. The key issues are additionality (is the new  
287 renewable source used to offset emissions additional to what would have been done anyway?)  
288 and allocation (the new renewable source could of course also be used to offset emissions from  
289 any other end users).

290

### 291 5.3 *Geothermal desalination*

292 Harvesting of geothermal energy to power desalination systems has a considerable advantage  
293 over renewable energy sources that can produce energy over part of a 24-hour daily cycle. A  
294 geothermal energy system can provide “base-load” power on a continuous basis and can do so  
295 for continuous, long-term time periods if properly designed. Coupling a geothermal energy  
296 source with electric power generation and desalination can produce the highest efficiency use of  
297 the resource (Figure 1). Using combined geothermal energy system with solar power for  
298 desalination of seawater has been proposed by Missimer et al. [33] to eliminate the necessity to  
299 develop thermal storage for nighttime operation of a purely solar-powered desalination system  
300 and to allow geothermal heat source regeneration.

301 There are three types of geothermal energy systems; wet rock/water flow (WR), natural dry  
302 steam, and hot dry rock (HDR). In a wet rock system, heat is extracted from either natural water  
303 flow from springs or from wells drilled into a hot-water aquifer. The cooled water is  
304 subsequently discharged to the environment or is re-injected into the groundwater system at  
305 some distance from the source to allow reheating and return into the system. It operates similar  
306 to the heat-pump systems used in building heating and cooling systems [34, 35] (Figure 2).  
307 Natural dry steam systems tend to occur near active volcanic activity where groundwater comes  
308 in contact with naturally heated rock and produces super-heated water under pressure in the  
309 subsurface. The drilling of a well into the aquifer containing the superheated water allows the  
310 harvesting of the steam with subsequent passage through a turbine electrical generation system  
311 (Figure 3). In HDR geothermal systems, heat is extracted by creating a man-made system of  
312 connected wells with artificial fractures used to connect an injected fluid from an injection well,  
313 through the fractures where heat is extracted, and ultimately to an extraction well where the  
314 superheated fluid is recovered [36-39]. The fluid used can be geothermal oil with a series of heat  
315 exchangers used at land surface to recover energy (low heat extraction) or low-salinity water can  
316 be superheated and flashed to steam at the well-head or near a turbine to generate electricity. The

317 steam flash system is more efficient in energy recovery compared to oil/heat exchange. The  
318 steam flash system is quite useful for coupling power generation with desalination.

319 The goal of geothermal energy development is to produce electric energy and desalinate water  
320 at the highest level of efficiency, using virtually all of the extracted latent heat in the process.  
321 The link between electric energy production and desalination is critical in order to maximize  
322 total heat conservation. Perhaps the most efficient system that can be used includes a closed  
323 loop, multiple-well system with water injected through one or more wells that are drilled with a  
324 horizontal offset and hydraulically fractured to produce a porous media (HDR system) (Figure  
325 4). The distilled water flows from the injection well through the labyrinth of fractures and is then  
326 pumped (or flows via natural head) into the recovery well. The water is maintained under  
327 pressure until reaching the proximity of a turbine system before which it is allowed to flash to  
328 steam, powering the production of electricity.

329 There are several configurations that can be used to couple geothermal energy with  
330 desalination processes. The first one is to couple geothermal energy production to MED. In this  
331 case the system can operate by using dry steam production from wells or from an HDR  
332 collection system. The steam is used to heat the raw water flowing into a standard MED unit.  
333 The second type would be to use geothermal energy to generate electrical power first and then  
334 run a stand-alone RO desalination plant. Also, a hybrid system could be developed that first  
335 generates electrical energy by using flashed steam from either a dry steam reservoir or a HDR  
336 energy extraction system. The steam vented through the turbines could be redirected for heating  
337 of the raw water going into an MED plant. The electrical energy could be used to power an  
338 onsite RO facility that combines flow with the MED plant to potable drinking water standards  
339 (Figure 5). This system would efficiently utilize the captured geothermal heat and would be  
340 particularly useful in treating high salinity waters, such as those of the Arabian Gulf and Red  
341 Sea.

342

## 343 **6. Contribution in developing renewable energy for desalination in KSA**

344 Our group's contribution is developing new desalination processes which can be driven by  
345 waste heat or SE, mainly AD, MD and forward osmosis (FO) technologies targeting energy  
346 consumption below 2 kWh/m<sup>3</sup>. An innovative hybrid approach was also explored which would  
347 combine solar and geothermal energy using an alternating 12-hour cycle to reduce the  
348 probability of depleting the heat source within the geothermal reservoir and provide the most

349 effective use of RE. The main advantage of these more sustainable technologies is that they are  
350 simple, operate at lower temperatures and pressures, and they can function with intermittent  
351 energy supply (variable loads) without additional operating modifications and energy storage,  
352 making them suitable for RE sources without requiring connection to the electrical grid.

353

### 354 *6.1 Adsorption desalination (AD)*

355 AD exploits the high affinity of water vapor onto an adsorbent caused by the double-bond  
356 surface forces that exist between the porous absorbent and an adsorbate such as the silica gel and  
357 water vapor. The pore diameters of the adsorbent range from 10 to 40 nm and the total pore  
358 surface area ranges from 600 to 800 m<sup>2</sup>/g. Figures 6 (a) and (b) show pictures of the silica gel  
359 and their typical pore distribution. The main advantage of using an adsorbent like silica gel is its  
360 ability to be re-generated by a low temperature heat source, typically from 55 to 85 °C (very  
361 suitable for RE use), and the high uptake rate of water vapor when exposed. When the adsorption  
362 phenomena are configured as a batch-operated cycle [40-42], an energy efficient cycle is  
363 obtained which can produce two useful effects with only one heat input, namely cooling and  
364 high-grade potable water; thus, the technology is more precisely defined as adsorption  
365 desalination with cooling (ADC). Over the past decade, Ng et al. [43-49] have successfully  
366 tested the silica gel and zeolite-based adsorption cycles (ADC), capturing low temperatures  
367 waste heat from industries as well as utilizing renewable solar and geothermal heat.

368 Figure 7 depicts the niche of the ADC cycle with respect to the temperature scale of heat  
369 sources. It produces both cooling and water by extracting the waste heat of exhaust or from the  
370 renewable solar and geothermal. The key point to note is that by utilizing the waste heat, it is  
371 “non-payable” because if such heat sources are untapped, they would be purged into ambient.  
372 Thus, the ADC technology is deemed to be environment-friendly and can reduce thermal and  
373 chemical pollution by re-cycling low temperature heat sources into useful effects.

374 A demonstration solar-AD facility has been constructed at KAUST and a life cycle assessment  
375 showed that a specific energy consumption of <1.5 kWh/m<sup>3</sup> is possible.

376 Figure 8 shows the schematic diagram of an advanced AD cycle, comprising an evaporator–  
377 condenser with stationary beds. The other major component in an ADC plant are (i) the feed  
378 water tank, (ii) two adsorber/desorber beds, (iii) the evaporator–condenser device, (iv) potable  
379 water collection tank and (v) the brine discharge tank. Each adsorber bed contains silica gels or  
380 zeolite which is packed around the tube-fin heat exchangers. During a batch-operated operation,

381 the reactor beds can be linked to the evaporator or the condenser during the half-cycle periods  
382 via a series of valves for the control of vapor and water flows. Consequently, an ADC cycle  
383 comprises two half-cycles (intervals vary from 200 to 700 s) and a switching interval (from 20 to  
384 40 s) in between which handles either the pre-heating or cooling of the exchangers. Based on our  
385 test experience, an ADC at a standard rating conditions ( $T_{\text{source}} = 85\text{ }^{\circ}\text{C}$ ,  $T_{\text{cooling}} = 30\text{ }^{\circ}\text{C}$  and  $T_{\text{chilled water}} = 7\text{ }^{\circ}\text{C}$ )  
386 could produce a specific cooling and water production capacities of 25 Rton and 12  
387  $\text{m}^3$  per day, respectively, per ton of silica gel. The evaporator of the ADC has been proven to  
388 handle a recovery of 80 % of seawater feed, which is much higher than any other conventional  
389 method. Table 3 illustrates the schedule of an ADC operation. Further details of such an  
390 emerging technology for low life-cycle cost and low temperature ADC desalination can be found  
391 in the published literature [40-49].

392

## 393 6.2 Membrane distillation (MD)

394 MD is an emerging desalination process that has been under investigation since the 1960s but  
395 has drawn more attention in the last decade. Details on the MD process have been widely  
396 reviewed [50-55]. MD is a thermally driven process that utilizes a hydrophobic, micro-porous  
397 membrane as a contactor to achieve separation by liquid-vapor equilibrium. The feed solution,  
398 after being heated, is brought into contact with the membrane which allows only the vapor to  
399 transport through the dry pores so that it condenses on the permeate side (coolant), as shown in  
400 Figure 9. This vapor is driven across the membrane by the difference in the partial vapor  
401 pressure maintained at the two sides of the membrane. A temperature difference of  $10^{\circ}\text{C}$  between  
402 the warm and cold streams is potentially enough to produce water [56].

403 MD holds high potential for several applications including water desalination application [51,  
404 55]. It holds the potential of being efficient and cost effective separation process that can utilize  
405 low-grade waste heat or RE (typical operating temperatures of  $60\text{-}70\text{ }^{\circ}\text{C}$ ) such as geothermal [57,  
406 58] or solar [52] energies, which are widely available in the region. One of the main advantages  
407 of MD is that the process performance is not highly affected by high feed salinity, as has been  
408 proven in bench scale [51, 59] and pilot scale [55] studies.

409 Different flat sheet and hollow fiber modules have been designed and fabricated by our group  
410 aiming to maximize the flux (product water). In addition to process engineering development,  
411 our MD research team has locally synthesized and fabricated hydrophobic membranes made of

412 different polymers (e.g. PVDF, PTFE, PP, fluorinated polytriazole materials) by different  
413 methods (e.g. phase inversion, electrospinning) [50, 51].

414 A water vapor flux of about 89 kg/m<sup>2</sup>h was obtained at feed and permeate inlet temperatures of  
415 80 °C and 20 °C, respectively, which is the highest reported flux obtained using real seawater as  
416 feed solution. A solar MD pilot plant is being designed using the optimized experimental data  
417 and will be tested at KAUST for long term operation using Red Sea water.

418

### 419 6.3 *Geothermal desalination*

420 A combined-cycle solar and geothermal powered AD process developed by our group was  
421 recently reported [33]. The major advantage of using this heat source configuration is that the  
422 daytime use of SE would not have to have a nighttime and/or cloudy period thermal energy  
423 storage component which can be quite costly to develop. Also, the development of HDR  
424 geothermal energy is less expensive because the depth of the wells can be less. The overall heat  
425 harvesting scheme can operate at lower temperatures and the heat sink will not deplete the heat  
426 reservoir because there is a recovery period each day.

427 Preliminary modeling conducted on the HDR system showed that the breakthrough of heat loss  
428 from an injection well to the recovery well will take greater than 8 days. This means that the  
429 HDR geothermal harvesting scheme can operate continuously for this period for the relatively  
430 shallow wells depth being considered at about 1,500 m below surface (Figure 10).

431

## 432 7. Conclusions

433 Water and energy resources are intertwined, within the water-energy nexus, and they no longer  
434 can be looked at as two separate issues and/or challenges. Both water and energy can be either  
435 non-renewable or renewable, depending on water and energy management and sources. While  
436 fossil fuels are clearly viewed as non-renewable, some water sources (e.g., fossil groundwater)  
437 can be viewed likewise. With rapid depletion of both resources at an alarming rate, there have  
438 been environmental ramifications, including GHGs, emissions causing global climate changes.  
439 Energy mixes including alternative resources, such as solar, wind and geothermal energies,  
440 should become a common practice. Their usage should become part of the energy demand,  
441 instead of fossil fuel resources; preserved for the future generations. Rather than using more  
442 energy for producing freshwater from seawater, more emphasis should be given on water reuse  
443 and recycling to minimize the stress on dwindling water supplies. Therefore, a more holistic and

444 universal approach is needed for resolving these critical issues. This paper has addressed some of  
445 the significant progress that have been occurring in recent years to attend these important issues  
446 through scientific research and policy changes. Now, utilities of all sizes are putting more  
447 emphasis on RE, such as solar, wind, geothermal, to incorporate them in water production.  
448 Although some are have already integrated in a medium to large scale setting, these have now  
449 become prime examples for reference. With the concept of keeping the energy usage in mind,  
450 many new technologies, such as FO, MD, AD and hybrid systems are now being developed.  
451 Some of these new, innovative technologies show that not only they are energy efficient in  
452 treatment, but also in incorporation of RE will make them even more efficient. These  
453 technologies use less energy and chemicals, resulting in producing less GHGs and climate  
454 change for the future. Therefore, more energy efficient and integrated green-technologies are  
455 needed to meet growing water and energy needs.

456

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Figure 1

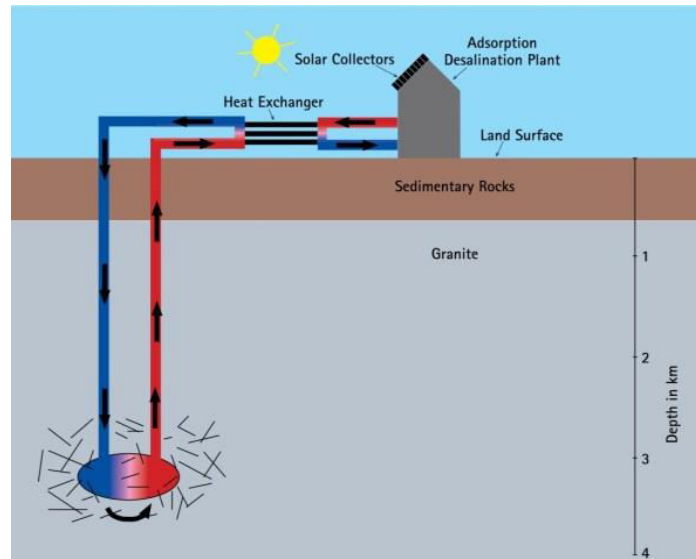


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Figure 2

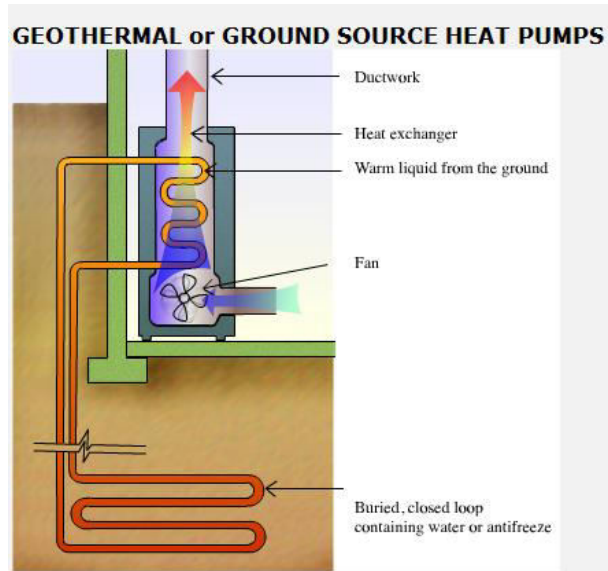


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Figure 3

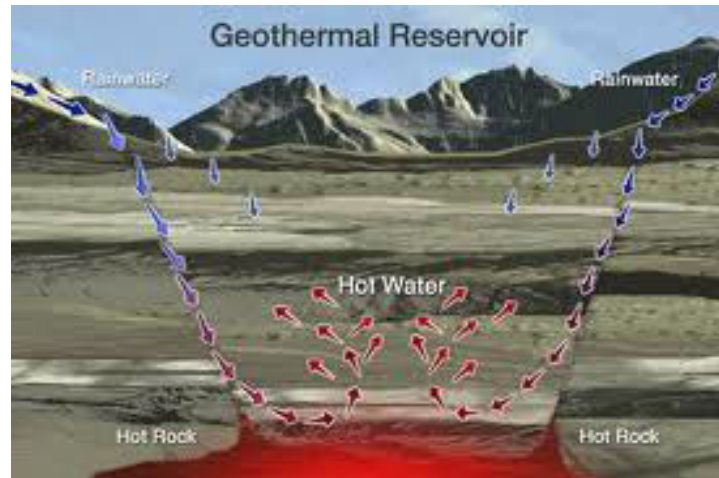


Figure 3: Dry steam can be collected by drilling a deep well into volcanically superheated water that flashed to steam at the wellhead (internet image).

Figure 4

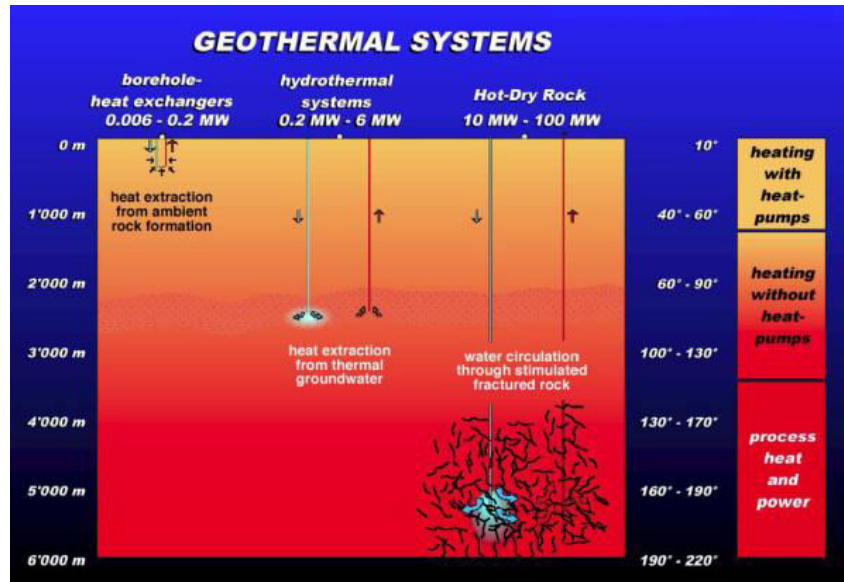


Figure 4: HDR geothermal collection systems can be engineered using hydraulic fracturing techniques to enhance fluid flow and allow better heat exchange.



Figure 5

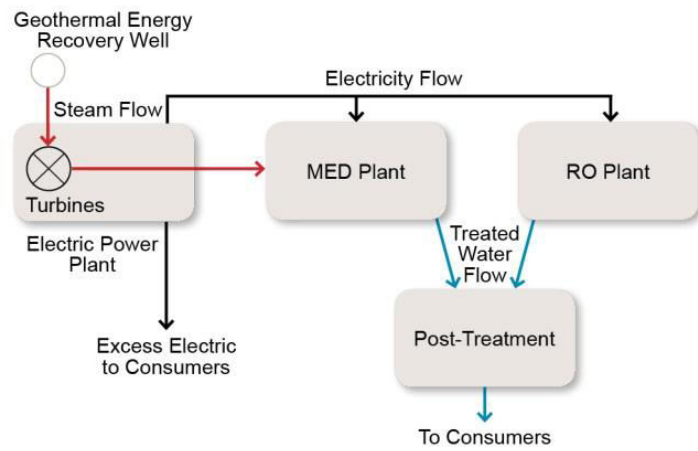


Figure 5: Coupled HDR geothermal energy collection-electric generation-desalination processes to efficiently cycle latent heat.

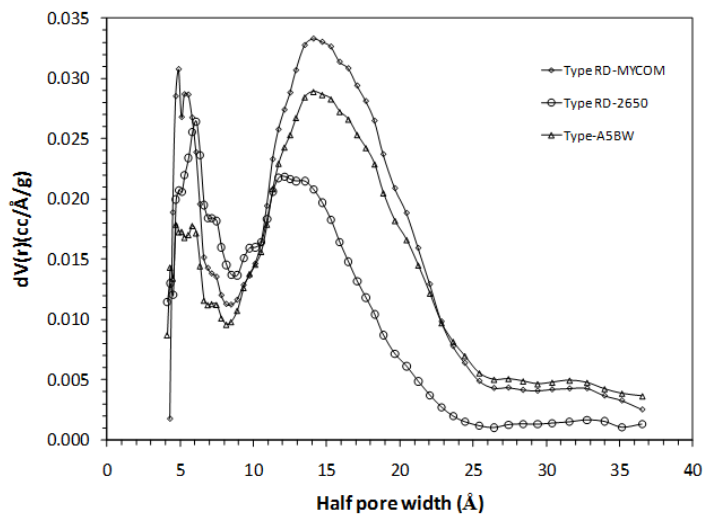
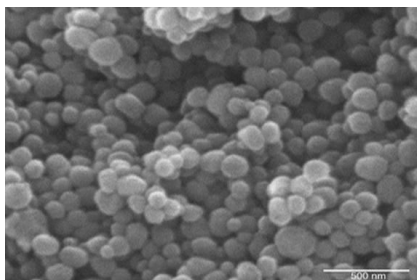


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Figure 7

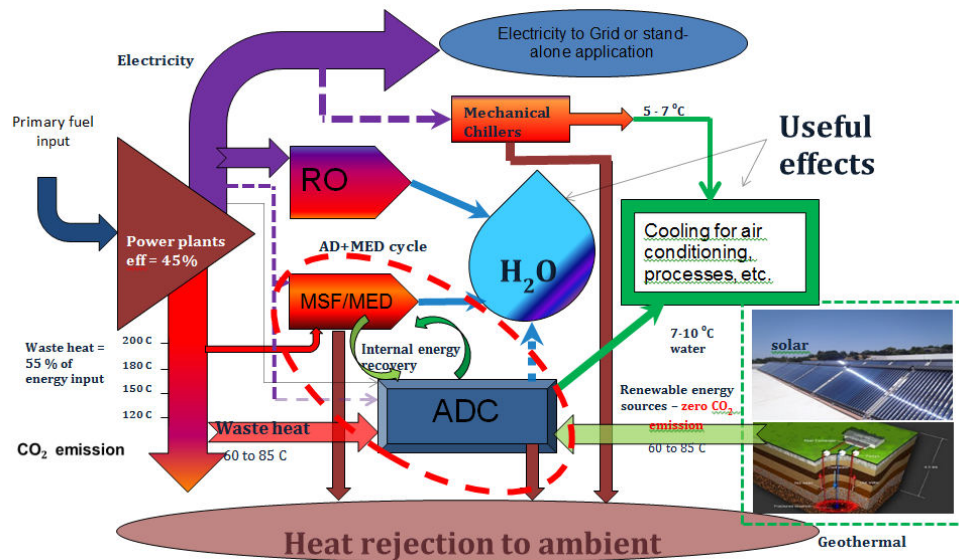
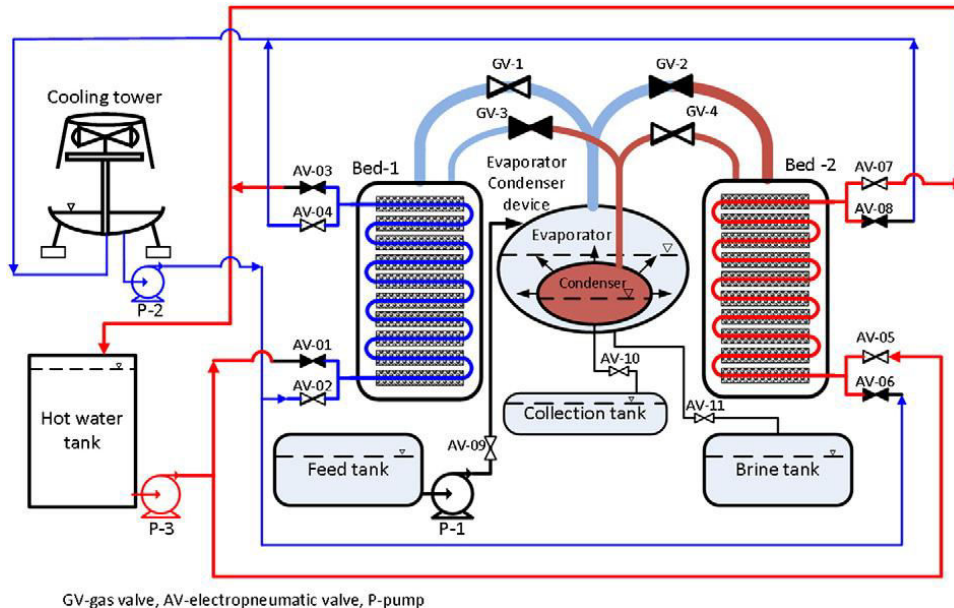


Figure 7: The niche of ADC technology for water and cooling production. It utilizes low temperature heat sources and is able to hybridize with conventional thermal desalination technologies.

Figure 8



GV-gas valve, AV-electropneumatic valve, P-pump

Figure 8: The schematic of an ADC with a 2-bed configuration. The specific cooling and water production capacities are 25 Rton and 12 m<sup>3</sup> per day, respectively, per ton of silica gel.

Figure 9

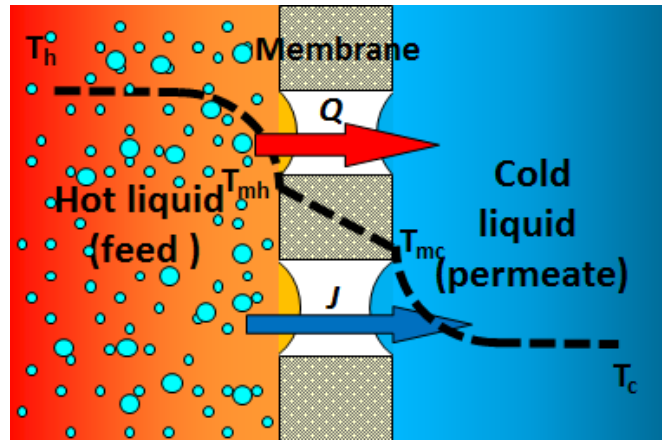


Figure 9: Principle of DCMD process.

Figure 10

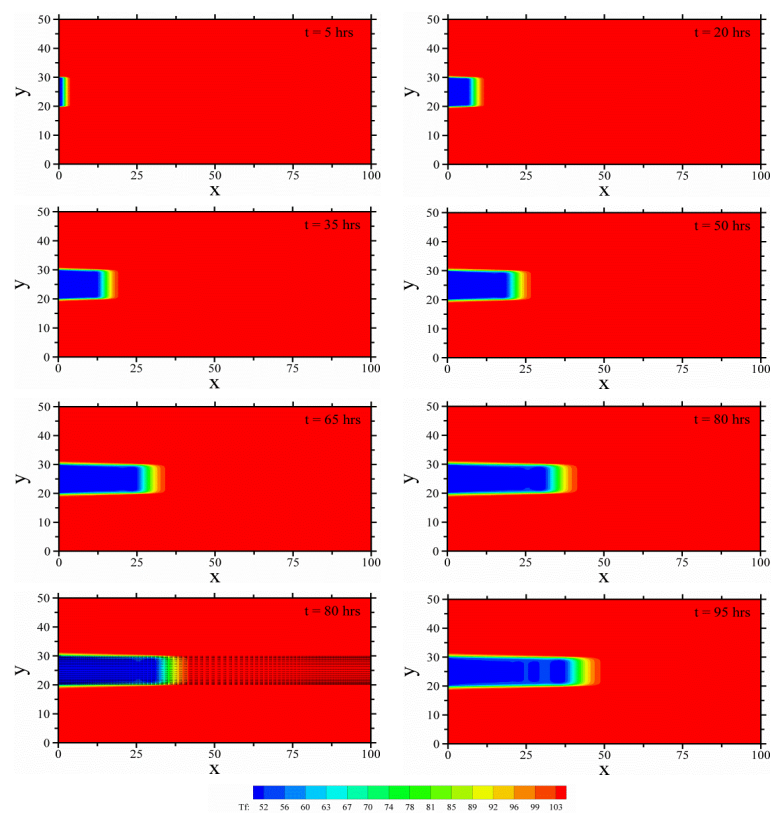


Figure 10: Model showing the migration of cool water from the injection well on the left side of the field to the recovery well on the right side of the field. The scale is in °C. Note that the cool water does not breakthrough to the recovery well for over 95 hours. The second graphic showing flow at t=80 hours includes the flow vectors [33].

Table 1: Desalination capacity in m<sup>3</sup>/day in KSA [12].

	MED	MSF	RO	Total
Gulf coast	833,844	3,140,459	923,020	4,897,323
Red Sea coast	324,780	2,446,478	1,707,012	4,478,270
Inland locations/inland	19,173	12,491	359,425	391,089
Total	1,177,797	5,599,428	2,989,457	9,766,682

Table 2: Energy demand of desalination processes and other water supply options [1, 15-17].

Water supply	Main energy form	Electrical energy [kWh/m <sup>3</sup> ]	Thermal energy [MJ/m <sup>3</sup> ]
BWRO	Electrical	0.5-3.0 [15]; 0.5-2.5 [1]	
SWRO	Electrical	2.5-7.0 [15]; 3-4 [1]	
MSF	Steam/thermal	3.0-5.0 [15]; 2.5-4 [1]	250-330 [15]
MED with TVC	Steam/thermal	1.5-2.5 [15]; 1.5-2 [1]	145-390 [15]
Surface water treatment	Electrical	0.2-0.4 [16, 17]	
Waste water reclamation	Electrical	0.5-1.0 [16, 17]	
Long-distance water transport *	Electrical	1.6-2.8 <sup>**</sup> / 12.0 <sup>***</sup>	

BWRO: Brackish water reverse osmosis

TVC: Thermal vapor compression

\* Depends on the transport distance and the elevation gap between source and destination, e.g., normal distribution costs are around 0.5-0.6 kWh/m<sup>3</sup>

\*\* Power required to convey surface water to San Diego, Los Angeles and Orange County [16]

\*\*\* Power required if water was conveyed to Perth via the Kimberley pipeline [18]



Table 3: A schedule of processes in a batch-operated ADC. The optimal intervals are dependent on the temperatures of heat source, cooling and chilled water.

Heat exchanger (-)	Cycle 1 (Time interval = 250-900s)	Switching 1 (Time interval = 20-45s)	Cycle 2 (Time interval = 250-900s)	Switching 2 (Time interval = 20-45s)
Bed-1	Adsorption	Pre-heating	Desorption	Pre-Cooling
Bed-2	Desorption	Pre-cooling	Adsorption	Pre-heating