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## Solid-state NMR Paramagnetic Relaxation Enhancement Immersion Depth Studies in Phospholipid Bilayers

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### Abstract

A new approach for determining the membrane immersion depth of a spin-labeled probe has been developed using paramagnetic relaxation enhancement (PRE) in solid-state NMR spectroscopy. A DOXYL spin label was placed at different sites of 1-palmitoyl-2-stearoyl-*sn*-glycero-3-phosphocholine (PSPC) phospholipid bilayers as paramagnetic moieties and the resulting enhancements of the longitudinal relaxation ( $T_1$ ) times of  $^{31}\text{P}$  nuclei on the surface of the bilayers were measured by a standard inversion recovery pulse sequence. The  $^{31}\text{P}$  NMR spin-lattice relaxation times decrease steadily as the DOXYL spin label moves closer to the surface as well as the concentration of the spin-labeled lipids increase. The enhanced relaxation vs. the position and concentration of spin-labels indicate that PRE induced by the DOXYL spin label are significant to determine longer distances over the whole range of the membrane depths. When these data were combined with estimated correlation times  $\tau_c$ , the  $r^{-6}$ -weighted, time-averaged distances between the spin labels and the  $^{31}\text{P}$  nuclei on the membrane surface were estimated. The application of using this solid-state NMR PRE approach coupled with site-directed spin labeling (SDSL) may be a powerful method for measuring membrane protein immersion depth.

### Keywords

solid-state NMR; paramagnetic relaxation enhancement; phospholipid membranes; spin-lattice relaxation

## 1. Introduction

One important piece of structural information of integral membrane proteins is the immersion depth within membrane, *i.e.*, how far a specific site of the protein is located from the membrane surface. Knowledge of the precise depth of a specific site or residue is essential in defining the overall topology and conformation of the proteins in the membranes, and helps in understanding the function and action mechanism of the system under investigation [1–5]. This is especially true since the full three-dimensional structures are not easily available for membrane proteins and macromolecular complexes.

A variety of biophysical techniques, with different intrinsic advantages and disadvantages, have been developed and employed to study the immersion depth of membrane proteins. Fluorescence techniques [6–10], FT-IR [11], and especially electron paramagnetic resonance

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(EPR) spectroscopy [12–21], including power saturation [17], ESEEM [19,20], have been widely used to obtain the immersion depth due to their relatively high sensitivity. The precision in determining the insertion depth using fluorescence and EPR spectroscopic methods is about 1.5–3 Å, owing to the size of either fluorescent or spin probes attached. Scattering techniques, mainly x-ray [22] and neutron diffraction [23,24], can yield a resolution of ~ 0.2 Å. A crucial issue for neutron diffraction is the detection limit for deuterium (i.e., the number of deuterons per lipid that can be detected) [24]. While most of the above techniques require sample labeling, deuteration for neutron diffraction, fluorophore-labeling for fluorescence quenching, or spin-labeling for EPR spectroscopy, NMR spectroscopy, either in the liquid-state [25–29] or in the solid-state [3,5,30–34], do not necessarily require isotopic labeling as long as reasonable sensitivity is achieved.

In standard solution NMR spectroscopy, the range of distance measured by NOE-based experiments is relatively short (~ 5 Å), owing to the weak interaction between nuclear dipolar magnetic moments. This interaction can be enhanced significantly by replacing the nuclear dipolar coupling with the much stronger electron-nuclear dipolar interaction, thus, resulting in paramagnetic relaxation enhancement (PRE) [35]. Previous theoretical and experimental studies have shown that significant PRE is expected for distances of ~ 5–30 Å [35,45]. PRE can serve as an ideal technique for determining the penetration depth in the membrane, since the thickness of a lipid bilayer is approximately ~ 30–50 Å.

Paramagnetic relaxation enhancement has been used in solution NMR spectroscopy to determine the immersion depth [25–28,36] and distances between paramagnetic centers [37, 38]. The paramagnetic agents can either be covalently linked to the site of interest [39,40] or freely soluble in the buffer surrounding the membrane environment, e.g.  $Mn^{2+}$  [41,42],  $O_2$  [43], lanthanide ions [44], or chelated paramagnetic metals [25,28]. One of the potential problems with free movable paramagnetic agents is the interaction between the agents and the proteins, which may lead to re-organization of the proteins in the membranes and change the native topology and conformation of the membrane proteins. Conversely, while PRE has been explored in solution NMR [45], it has been underutilized in solid-state NMR spectroscopy, whereas PRE is more pronounced in solid-state than in liquids [46]. Since solid-state NMR spectroscopy can provide powerful structural and dynamic information for investigating proteins in membranes, it is thus desirable to explore more fully the intrinsic benefits of PRE with solid-state NMR [41,46–50].

One of our long-term research goals is to fully explore the unique advantages of both solid-state NMR and EPR spectroscopy and to develop new structural biology tools for studying membrane proteins [51–53]. Since  $^{31}P$  nuclei are abundant at the surface of phospholipid membranes and are very NMR sensitive, we reasoned that site-directed spin-labeled (SDSL) proteins might be ideally used to determine the distances between the  $^{31}P$  nuclei and the spin labels by  $^{31}P$  solid-state NMR PRE, and one such application would be the determination of membrane insertion depth. In the work presented here, we employed a series of doxyl labeled 1-palmitoyl-2-stearoyl-*sn*-glycero-3-phosphocholine (n-DOXYL-PSPC, where n = 5, 10, 14, Fig. 1A) embedded into PSPC phospholipid bilayers.  $^{31}P$  longitudinal relaxation enhancement was determined to obtain qualitative measurements of the immersion depth of the spin-labels in the lipid bilayers (Fig. 1B).

## 2. Methods

All phospholipids were purchased from Avanti Polar Lipids, Inc. (Alabaster, AL) and used without further purification. 1-palmitoyl-2-stearoyl-*sn*-glycero-3-phosphocholine (PSPC), 1-palmitoyl-2-(n-doxylstearoyl)-*sn*-glycero-3-phosphocholine (n-DOXYL-PSPC, where n = 5, 10, or 14) were dissolved in chloroform and stored at -20 C° prior to use.

Multilamellar vesicles (MLVs) of pure PSPC or PSPC/n-DOXYL-PSPC mixtures in different molar ratios were prepared according to our previously reported procedure [54]. A total amount of ~70 mg of PSPC lipids was dissolved in chloroform and appropriate amounts of n-DOXYL-PSPC were mixed to obtain different molar ratios of spin-labels. The chloroform solvent was removed slowly under a steady stream of pure N<sub>2</sub> gas and the samples were dried overnight in a vacuum desiccator. The resulting mixtures were fully hydrated with 190 μL HEPES buffer (30 mM HEPES, 20 mM NaCl, pH 7.4). The hydrated samples went through 5–10 freeze-thawing cycles to increase the homogeneity of the vesicles. The pellet (60–100 mg) was transferred to a 4 mm MAS NMR rotor for <sup>31</sup>P NMR spectroscopic studies.

All NMR data were collected on a Bruker AVANCE 500 MHz wide-bore NMR spectrometer equipped with a CP-MAS probe (Bruker Biospin, Rheinstetten, Germany). <sup>31</sup>P longitudinal relaxation times (T<sub>1</sub>) were measured using a standard one-dimensional direct phosphorous detection inversion-recovery pulse sequence (180°-T- 90°- acquire), with a pre-acquisition delay of 10 s, and 14 relaxation delay D1 values ranging from 0.2 ms to 7 s. The sample was spinning at a frequency of 4 kHz at the magic angle. T<sub>1</sub> values were calculated by fitting the <sup>31</sup>P MAS isotropic peak intensities to a single exponential function I(t) = I(0) + P\*exp (-t/ T<sub>1</sub>) using Topspin 2.0 (Bruker Biospin).

The paramagnetic relaxation enhancement (PRE, described as Γ<sub>1</sub> in the following sections) is determined by subtracting the longitudinal relaxation rate of the membrane system in the absence of paramagnetic labeling (R<sub>1(dia)</sub> or 1/T<sub>1(dia)</sub>) from the relaxation rate in the presence of spin-labeled species (R<sub>1(para)</sub>, or 1/T<sub>1(para)</sub>) according to Eq. 1

$$\Gamma_1 = R_{1(para)} - R_{1(dia)} = \frac{1}{T_{1(para)}} - \frac{1}{T_{1(dia)}} \tag{1}$$

PRE Γ<sub>1</sub> can be calculated according to the modified Solomon-Bloembergen equations (Eq. 2) [37,45].

$$\Gamma_1 = \frac{2}{5} \left( \frac{\mu_0}{4\pi} \right)^2 \gamma_i^2 g^2 \mu_B^2 S(S+1) J_{SB}(\omega_r) \tag{2}$$

Where γ<sub>r</sub> is the gyromagnetic ratio of <sup>31</sup>P, g is the electron g-factor, μ<sub>0</sub> is the magnetic permeability of vacuum, μ<sub>B</sub> is the electron Bohr magneton, S is the electron spin quantum number, ω<sub>r</sub>/2π is the Larmor precession frequency of <sup>31</sup>P, J<sub>SB</sub>(ω<sub>r</sub>) is the generalized spectral density for the reduced correlation function, which is a Lorentzian function of ω:

$$J_{SB}(\omega_r) = r^{-6} \frac{\tau_c}{1 + (\omega\tau_c)^2} \tag{3}$$

Where r is the distance between the unpaired electron (approximately localized on the nitrogen atom of the nitroxide DOXYL spin label) and the observed nuclear spins (here the <sup>31</sup>P nucleus), τ<sub>c</sub> is the correlation time of the electron – <sup>31</sup>P dipolar interaction. The general dependence of the distance r between the spin label and the <sup>31</sup>P nucleus on the correlation time is shown on Fig. 2. For the distance calculations an approximation was made that τ<sub>c</sub> was put to be equal to the global correlation time of the spin-labeled lipids.

In this paper, the items  $\frac{2}{5} \left( \frac{\mu_0}{4\pi} \right)^2 \gamma_i^2 g^2 \mu_B^2 S(S+1)$  are combined into a constant K. For <sup>31</sup>P nuclei, K = 1.21 × 10<sup>-32</sup> cm<sup>6</sup> s<sup>-2</sup>.

Thus, Eq. 2 can be re-written as:

$$r = \left[ \frac{K}{\Gamma_1} \frac{\tau_c}{1 + (\omega\tau_c)^2} \right]^{1/6} \quad (4)$$

Now, the DOXYL concentration should be taken into account. This can be done by integrating  $^{31}\text{P}$  relaxation signals from all  $^{31}\text{P}$  atoms of the lipid surface. The integration area can be estimated as proportional to the square root of the DOXYL-labeled lipids/overall lipids ratio (i.e. the DOXYL-label concentration). Thus the inversion recovery signal can be described as follows:

$$S = \int_0^{R_{lim}} 2\pi r \rho \left( y_0 + A \exp \left( -\frac{t}{T_{pspc}} - \frac{K}{(r^2 + d^2)^3} \frac{t \tau_c}{1 + (\omega\tau_c)^2} \right) \right) dr \quad (5)$$

where  $\rho$  is DOXYL concentration,  $y_0$  and  $A$  are the level offset and the amplitude of the measured inversion recovery curve,  $T_{pspc}$  is the relaxation time of pure PSPC lipids,  $r$ ,  $d$  and  $R_{lim}$  are defined as in Fig. 5. This equation has an analytical solution:

$$S = \pi (R_{lim}^2 + d^2) \rho \left( y_0 + A \exp \left( -\frac{t}{T_{pspc}} - \frac{K}{(R_{lim}^2 + d^2)^3} \frac{t \tau_c}{1 + (\omega\tau_c)^2} \right) - A \exp \left( -\frac{t}{T_{pspc}} \left( \frac{K}{(R_{lim}^2 + d^2)^3} \frac{t \tau_c}{1 + (\omega\tau_c)^2} \right)^{\frac{1}{3}} \right) \Gamma \left( \frac{2}{3}, \frac{K}{(R_{lim}^2 + d^2)^3} \frac{t \tau_c}{1 + (\omega\tau_c)^2} \right) \right) \quad (6)$$

where  $\Gamma$  is upper incomplete gamma function. Fitting this equation to the experimental data gives a relaxation behavior that accounts for  $^{31}\text{P}$  – DOXYL-label distance distribution and allows a more accurate calculation of the DOXYL depth. Further we refer to the relaxation behavior described by (6) as “geometry weighted relaxation”.

### 3. Results

PSPC vesicles with spin-label located at three different depths (Fig. 1) of the lipid bilayer was added respectively to the control pure PSPC lipid bilayer and the  $^{31}\text{P}$   $T_1$  time values were determined using the inversion-recovery pulse sequence. The inversion recovery data and the corresponding  $T_1$  fits for the n-DOXYL-PSPC/PSPC (n=5, 10, 14) along with that for the pure PSPC lipid vesicles are shown in Fig. 3. In all cases, the decay resulting from the relaxation phenomenon can be fitted satisfactorily with a single exponential behavior, suggesting that there is a single predominant contribution to the relaxation process. However, fitting with geometry weighted relaxation model gives slightly different values. As clearly observed in Fig. 3, adding the paramagnetic agents of n-DOXYL PSPC lipids induced faster  $^{31}\text{P}$   $T_1$  relaxation. As expected the relaxation enhancement depends upon the position of the DOXYL spin label with respect to the  $^{31}\text{P}$  nuclei on the membrane surface.

The  $^{31}\text{P}$   $T_1$  values for the systems studied are summarized in Table 1. To address the paramagnetic relaxation enhancement of the nuclear spin-lattice relaxation rates that results from intermolecular magnetic dipole-dipole interactions, and to find the optimal concentration of spin-labels to lipid, a titration of the 5-DOXYL-PSPC vesicles was conducted, and the corresponding  $^{31}\text{P}$   $T_1$  relaxation times were measured. Plotting  $T_1$  relaxation times as a function of the molar ratio of the spin-labels yields a straight line whose slope is the paramagnetic

relaxation enhancement (Fig. 4). As expected, the  $T_1$  values depend linearly on the relative molar fraction of the spin-label concentration in MLVs [56].

We evaluated the contribution of the intermolecular magnetic dipole-dipole interactions to the PRE at low concentration of paramagnetic species by estimating the distance between the  $^{31}\text{P}$  nuclei and the nearby spin label (Fig. 5). The surface area of per PSPC molecule is  $67 \text{ \AA}^2$  [57] which results in a separation of  $R = 8.2 \text{ \AA}$  between the two head groups. At the concentration of 5% mol n-DOXYL PSPC, the next nearest spin label will be about  $37 \text{ \AA}$  away from the  $^{31}\text{P}$  nucleus (Fig. 5) and thus, due to the strong distance dependence, the PRE effect can be seen as a single nucleus-spin label process for the distances of interest in the range of  $10\text{--}17 \text{ \AA}$ . It should be noted that in general one does not need to carry out a full titration with spin-labels. Instead, acquiring relaxation times in the absence and presence of a significant concentration of the spin-labeled samples should suffice.

#### 4. Discussion

The membrane immersion depth, or penetration depth, is not well defined in the literature. In fluorescence studies developed by London [6], often the distance from the spin-label to the bilayer center was estimated. In EPR spectroscopy studies, generally another paramagnetic agent needs to be added to measure the distance between the spins and hence the immersion depth. In a neutron diffraction study [24], distance of deuterated amino acids from the bilayer center were determined. In ESEEM spectroscopy, the head groups of lipids were deuterated and the distance between deuterium and the spin, which is placed at different positions in the membrane, were determined [20]. In previous NMR PRE studies in micelles, the distance between the nucleus and paramagnetic agents such as  $\text{Mn}^{2+}$  on the surface of lipid bilayer [42], chelated paramagnetic metal Gd(DTPA-BMA) on the surface of the micelle [25,28,56] were used.

$^{31}\text{P}$ , with a 100% natural abundance and an intermediate gyromagnetic ratio, are distributed uniformly on the surface of the lipid bilayer and can be used as an ideal reference for immersion depth measurements by NMR spectroscopy. The marked enhancement of the relaxation efficiency observed in the presence of the paramagnetic species (Fig. 3, Fig. 4 and Table 1) is the result of the magnetic moment of the unpaired electron's being almost 1000 times higher than that of the  $^{31}\text{P}$  nucleus. The dipolar interaction between the electron and nuclear spins is much stronger than the dipolar interaction between two nuclei, and its strength depends on the distance and is not influenced by the medium in between them; therefore, such measurements are potentially very precise. This approach also eliminates the necessity for addition of paramagnetic agents on the surface of the membrane or deuteration of the headgroups of the lipids.

The paramagnetic DOXYL probe used in this paper is a nitroxyl ring attached rigidly at a specific carbon position on the chain of PSPC lipids. Previous studies have shown that DOXYL-PSPC is rather non-perturbing for lipid membranes [58].

As shown in Fig 3, introduction of the spin-labels induced significant PRE effect as observed in the changes of the  $^{31}\text{P} T_1$  times and the resulting PRE show clear dependence on the distances between the spin labels and the  $^{31}\text{P}$  nuclei. This PRE provides a relative spectroscopic ruler to measure the depth of a specific site in the membranes. PRE also depends upon the concentration of the spin labels (Fig. 4) and the correlation time  $\tau_c$ . The higher concentration of spin labels, the more significant PRE is. The graph is linear over the concentration range shown in Figure 4. At higher concentrations, this graph may not be linear because of lipid demixing. Our single pulse  $^{31}\text{P}$  NMR data indicate that 5% mol of spin labels is enough for inducing a significant PRE, while keeping the structure of the membrane intact.

The estimated PRE distances in Table 1 were compared with the available literature distances [12]. The exact distance between  $^{31}\text{P}$  and spin labeled lipid membranes is not available in the literature. According to force field and density functional theory (DFT) calculations, expected distances for an all-trans conformation of the hydrocarbon chain are 10 Å for 5-DOXYL-PSPC in the gas phase [59]. By using fluorescence quenching, the distances from the hydrophobic center of the bilayers to the spin-label were estimated to be 12.2 Å for the 5-DOXYL PSPC [6], which gives a distance of  $\sim 11$  Å between the 5-DOXYL and the membrane surface, assuming the thickness of a PSPC bilayer to be 46 Å [60]. Our corresponding calculated distance of 10.6 Å for 5-DOXYL-PSPC is in a reasonable range.

When the spin-label is moving towards the center of the PSPC bilayer,  $^{31}\text{P}$   $T_1$  relaxation times and corresponding distances increase (see Table 1). It has been shown that correlation times vary with the positions in DOXYL-labeled stearic acid (6.5 ns for 5-SDA, 12.4 ns for 10-SDA) [61]. We used an estimated correlation time  $\tau_c$  in calculating the distance from the PRE data of 1 ns. Because the distance depends on the sixth-root of the correlation time, the error in distance  $r$  introduced by even large errors in correlation time  $\tau_c$  is relatively small. For instance, a relative error of 50% in the correlation time for a given site would produce an error of about 9% in the measured distance, or approximately  $\pm 1.5$  Å for a distance of 15 Å [55]. The calculation yields the following values for the  $^{31}\text{P}$ -spin label distances: 12.6, 13.3 and 16.6 Å, for 5-, 10- and 14-DOXYL PSPC respectively for a single exponential fit and 10.6, 12.3 and 16.0 for a geometry weighted relaxation fit which are more physically appropriate. This difference clearly shows the necessity of taking into account the relaxation signals from all nearby lipids.

An obvious improvement of this approach is to conveniently determine the correlation time with EPR spectroscopy [62]. Another source of error is the relaxation curves fitting procedure. Since  $\Gamma_1$  is in the denominator and is a difference between two relaxation rates a careful estimation of relaxation times is needed especially for the long distances as a small error in  $\Gamma_1$  calculation can lead to huge errors in distance calculations.

Several assumptions were made in our model of membrane immersion depth determination by  $^{31}\text{P}$  PRE  $T_1$  of the phospholipids with spin labeling. The electron- $^{31}\text{P}$  distance in fact is the most probable distance of a paramagnetic center to the planar surface of phospholipid membranes because the  $^{31}\text{P}$  nuclei are distributed uniformly in a plane of lipid membranes. It includes distance averaging due to molecular dynamics, lipid chain tilt angles, etc. In membrane peptide/protein systems, the molecular dynamics of the lipid will be much less significant, which should lead to an even more accurate measurement of immersion depth of proteins/peptides with angstrom resolution. A more detailed theory of PRE in lipids is being prepared in our group that will account for spin label concentration, different correlation time for each spin-label position and molecular dynamics.

With all limitations described above, PRE-based insertion depth measurements have obviously many advantages and when compared to other techniques. There is no need to have high-quality crystals and no limitation on molar mass. In EPR spectroscopy, another paramagnetic agent needs to be added to measure the distance between the spins and hence the immersion depth. The experiment can be conducted in the absence of any protein, in the presence of spin labeled protein, or proteins without spin labeling, and thus determine if the protein position and the overall structure of the bilayers changes in the presence of the label. The solid-state NMR method has the advantage of not requiring the coupling of an extrinsic spin probe to the protein.

In summary, we provided a qualitative estimation of distances between paramagnetic centers embedded in lipid bilayers and the  $^{31}\text{P}$  nuclei membrane surface and evaluated its feasibility in determining the immersion depth. Often for large macromolecular complexes in the



membrane, a few critical structural features including the location of a specific site can provide many insights on the function and mechanism of action. Combined with EPR spectroscopy this technique looks very promising in providing site-specific distance determination in complex systems where it is very challenging for many other techniques. We are extending the methodology developed here to determine the immersion depth of membrane proteins via SDSL coupled with  $^{31}\text{P}$  MAS solid-state NMR spectroscopy.

## Abbreviations used

<b>PSPC</b>	1-palmitoyl-2-stearoyl- <i>sn</i> -glycero-3-phosphocholine
<b>n-DOXYL-PSPC</b>	1-palmitoyl-2-stearoyl-(n-doxyl)- <i>sn</i> -glycero-3-phosphocholine
<b>ESEEM</b>	Electron Spin Echo Envelope Modulation
<b>MAS</b>	Magic-Angle Spinning
<b>PRE</b>	paramagnetic relaxation enhancement

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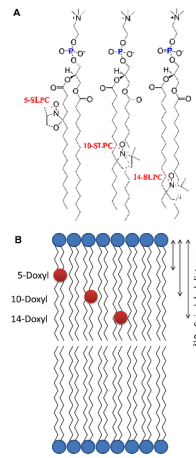
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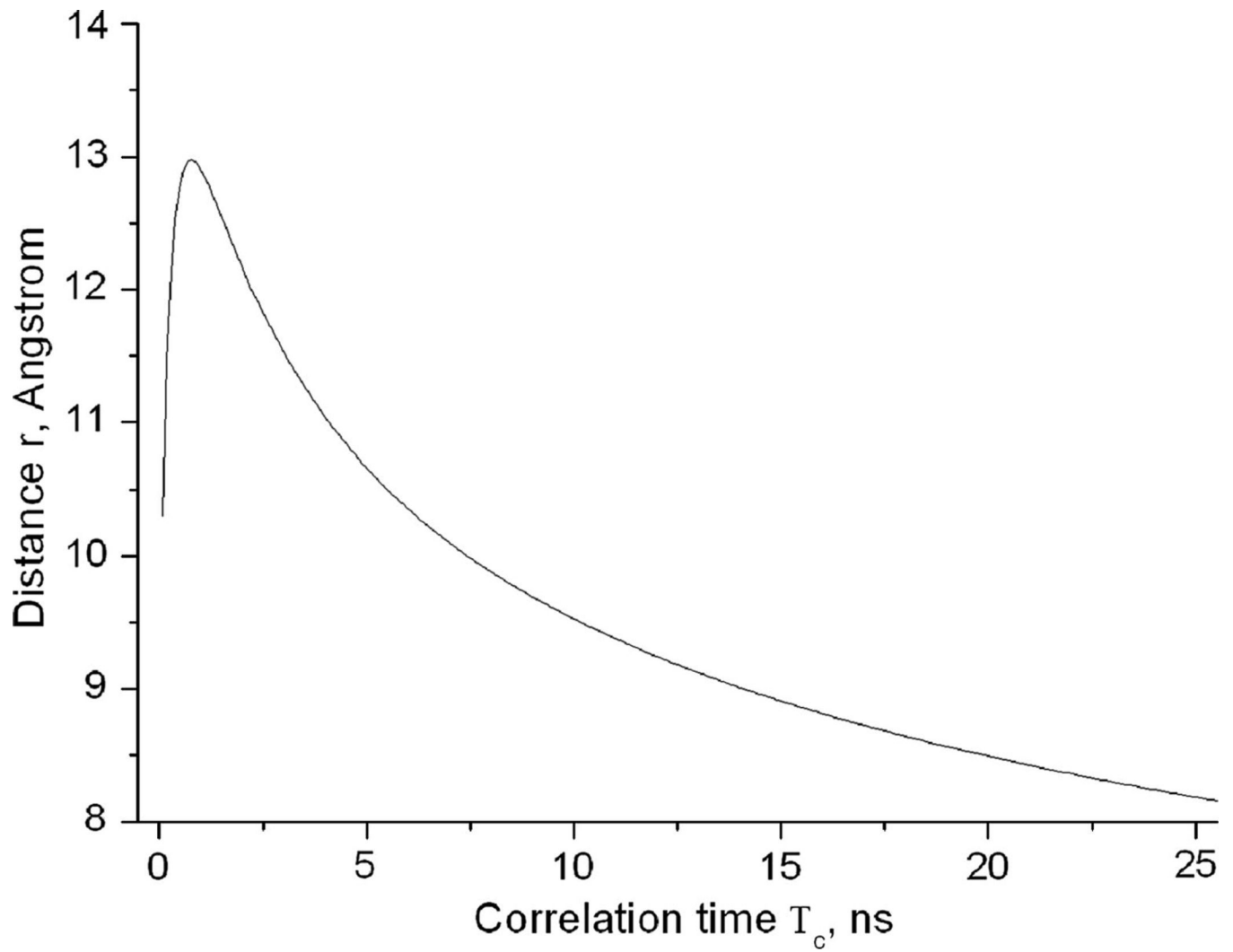


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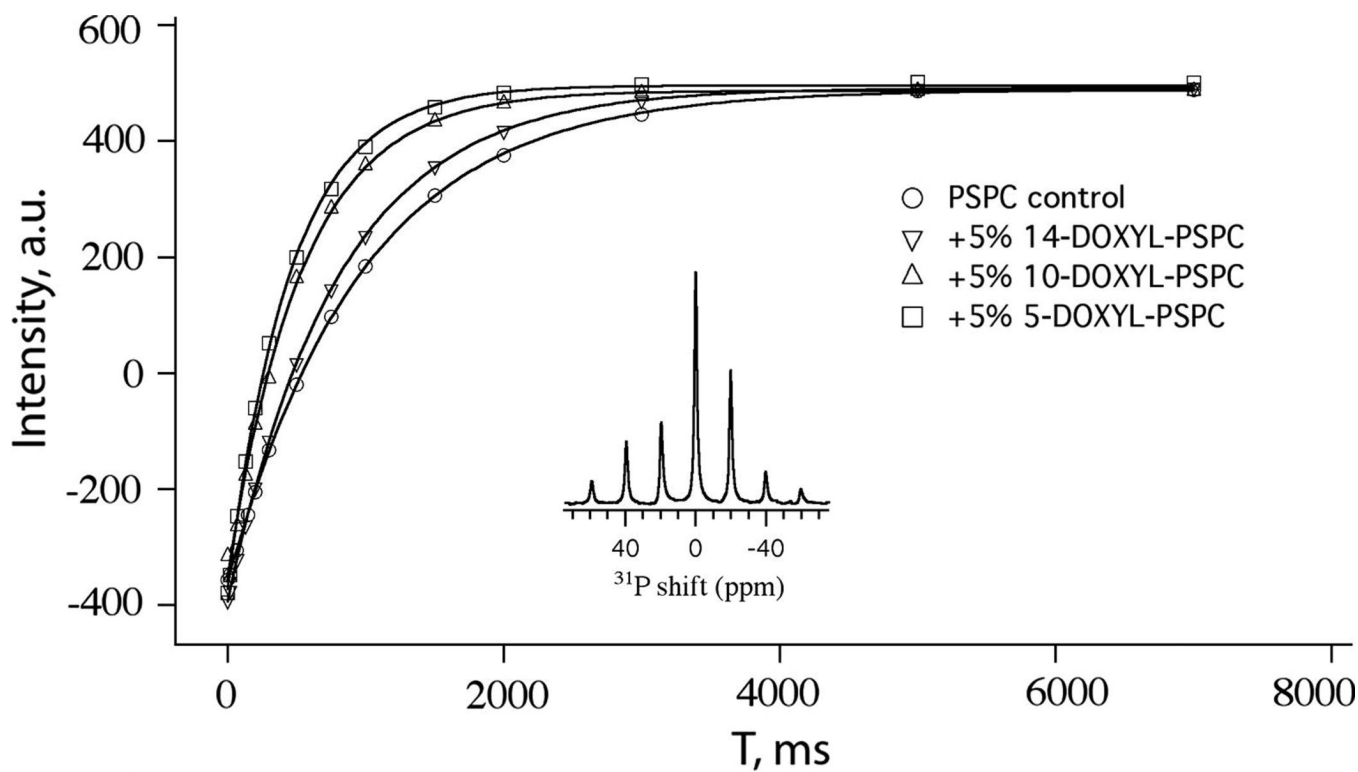
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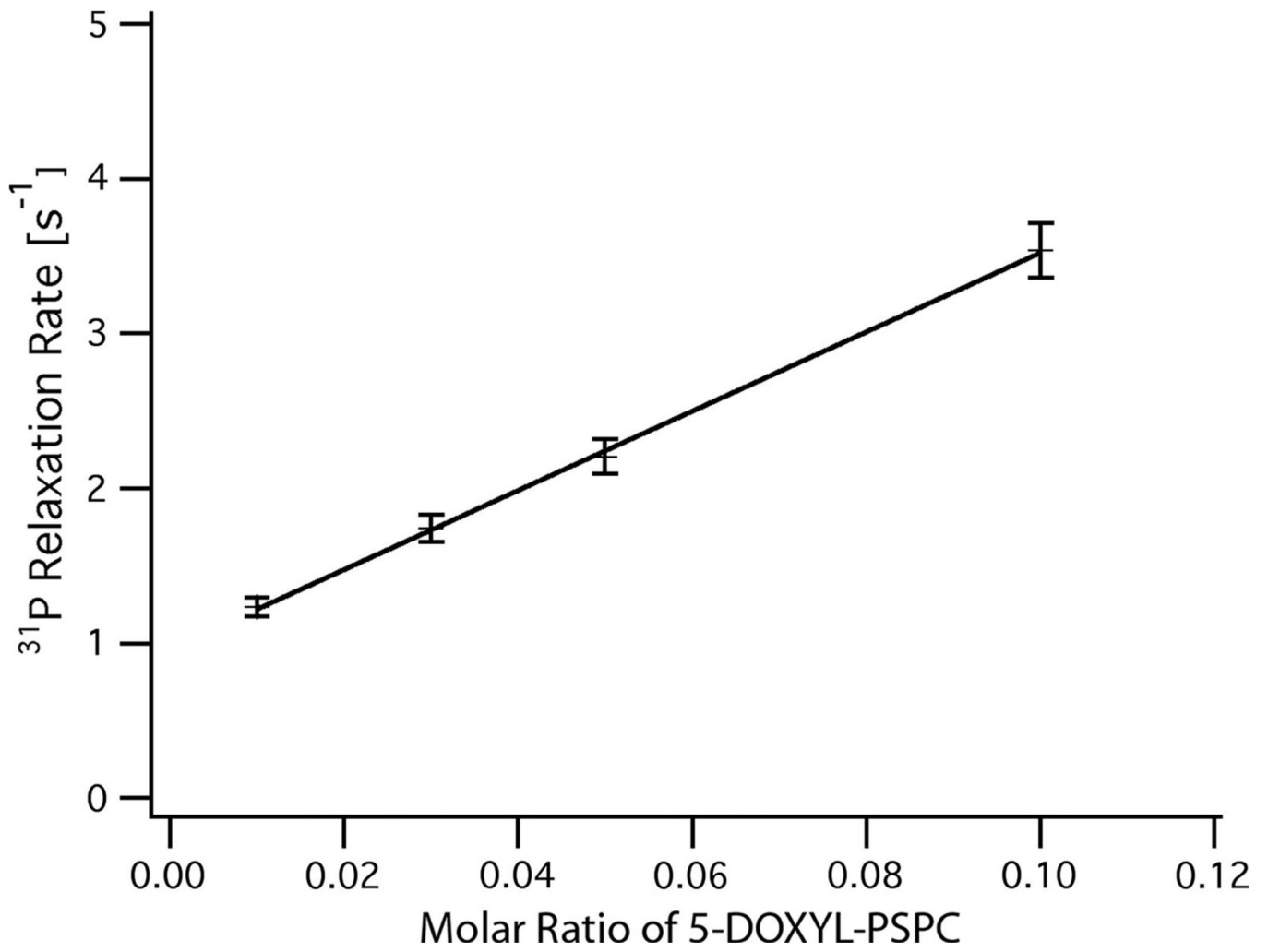
**Fig. 1.** (A) Chemical structures of n-(4,4-dimethyl-oxazolidine-N-oxyl)-palmitoyl-2-stearoyl-*sn*-glycero-3-phosphocholine (n-DOXYL-PSPC, where n=5, 10, 14) and (B) the phospholipid bilayer model. The membrane immersion depth is defined as the shortest distance between the <sup>31</sup>P and the spin-label DOXYL.



**Fig. 2.** Spin label -  $^{31}\text{P}$  distance vs. the correlation time  $\tau_c$ . Calculated distance changes only in the range of 8 to 13 Å whereas correlation time lies in the range from 0.1 to 25 ns.

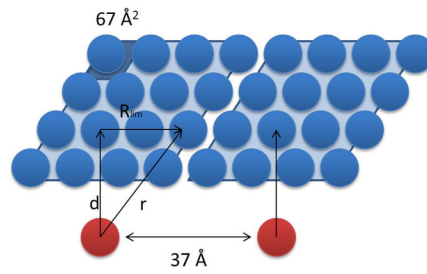


**Fig. 3.**  $^{31}\text{P}$  T<sub>1</sub> vs. the position of DOXYL on PSPC at 5% (molar ratio) of spin-labels. The signal intensity profiles of  $^{31}\text{P}$  are plotted as a function of the time delay in inversion recovery pulse sequence under control and in the presence of n-DOXYL PSPC. The insert is the typical  $^{31}\text{P}$  MAS NMR spectrum used to measure  $^{31}\text{P}$  T<sub>1</sub> by using inversion recovery, with sample spinning speed of 4 KHz at the magic angle.



**Fig. 4.** Plot of  $^{31}\text{P}$   $T_1$  times vs. the concentration of 5-DOXYL-PSPC. A liner relationship (for details see text).





**Fig. 5.** The concept model of the distance and the immersion depth (not drawn to scale). At a concentration of 5 mol % spin labeled lipid, there is one spin label to relax nearby  $20 \text{ }^{31}\text{P}$  nuclei. According to the area surface and the concentration, the distance between the observed  $^{31}\text{P}$  and another nearby spin label would be at least  $\sim 37 \text{ \AA}$  away (see text).

**Table 1**

<sup>31</sup>P T<sub>1</sub> values and the calculated distances (r).

Samples	T <sub>1</sub> (ms) <sup>*</sup>	Distance r (Å)				
		NMR <sup>‡</sup>		EPR <sup>12</sup>	Fluorescence	X-ray
		ExpFit <sup>*</sup>	GeomFit <sup>**</sup>			
PSPC	980					
5-DOXYL	463	12.6	10.6±0.1	13.6	10.9	8.1
10-DOXYL	537	13.3	12.3±0.3	17.6		14.5
14-DOXYL	806	16.6	16.0±0.5			

Note. <sup>31</sup>P T<sub>1</sub> were measured by using inversion recovery pulse sequence at 35°C with spinning at 4 KHz at magic-angle.

<sup>‡</sup>Distance calculated in this paper, using τ<sub>c</sub> = 1 ns (see text).

<sup>\*</sup> ExpFit – one exponential fitting

<sup>\*\*</sup> GeomFit – geometry weighted exponential fitting